

Supporting Information for

Ternary MOFs-Based Redox Active Sites Enabled 3D-on-2D

Nanoarchitected Battery-Type Electrodes for High-Energy Density

Supercapatteries

Goli Nagaraju^{1,2}, S. Chandra Sekhar¹, Bhimanaboina Ramulu¹, Sk. Khaja Hussain^{1,2},
D. Narsimulu¹, Jae Su Yu^{1,*}

¹Institute for Wearable Convergence Electronics, Department of Electronic Engineering, Kyung Hee University, 1732 Deogyong-daero, Yongin-si, Gyeonggi-do 17104, Republic of Korea

²Department of Chemical Engineering, College of Engineering, Kyung Hee University, 1732 Deogyong-daero, Gihung-gu, Yongin-si, Gyeonggi-do 17104, Republic of Korea

*Corresponding author. E-mail: jsyu@khu.ac.kr (Jae Su Yu)

Supplementary Figures and Tables

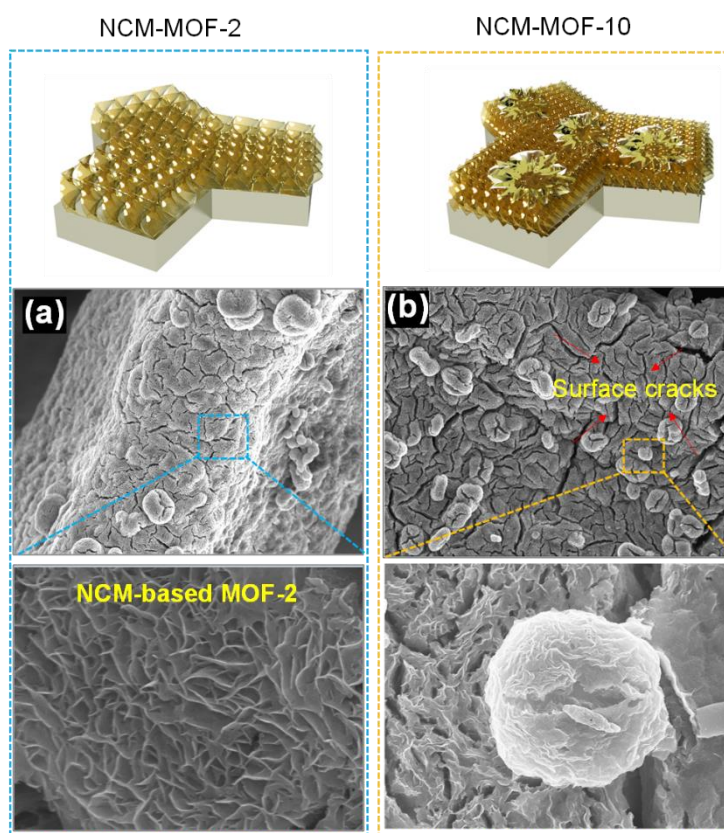


Fig. S1 Low- and high-magnification SEM images of the (a) NCM-based MOF-2/Ni foam and (b) NCM-based MOF-10/Ni foam electrodes

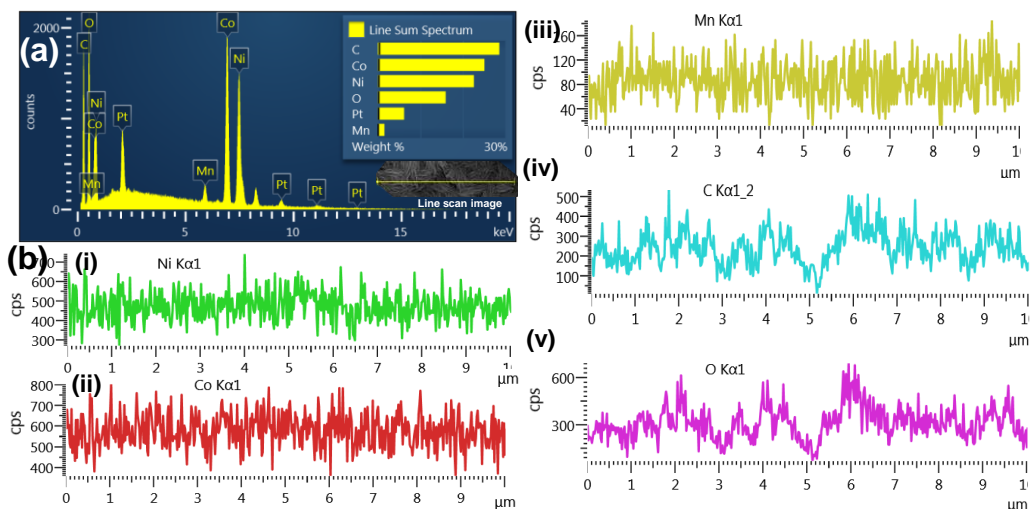


Fig. S2 EDX spectrum and line-scan spectra of individual elements (Ni, Co, Mn, C and O) for the NCM-based MOF-5/Ni foam

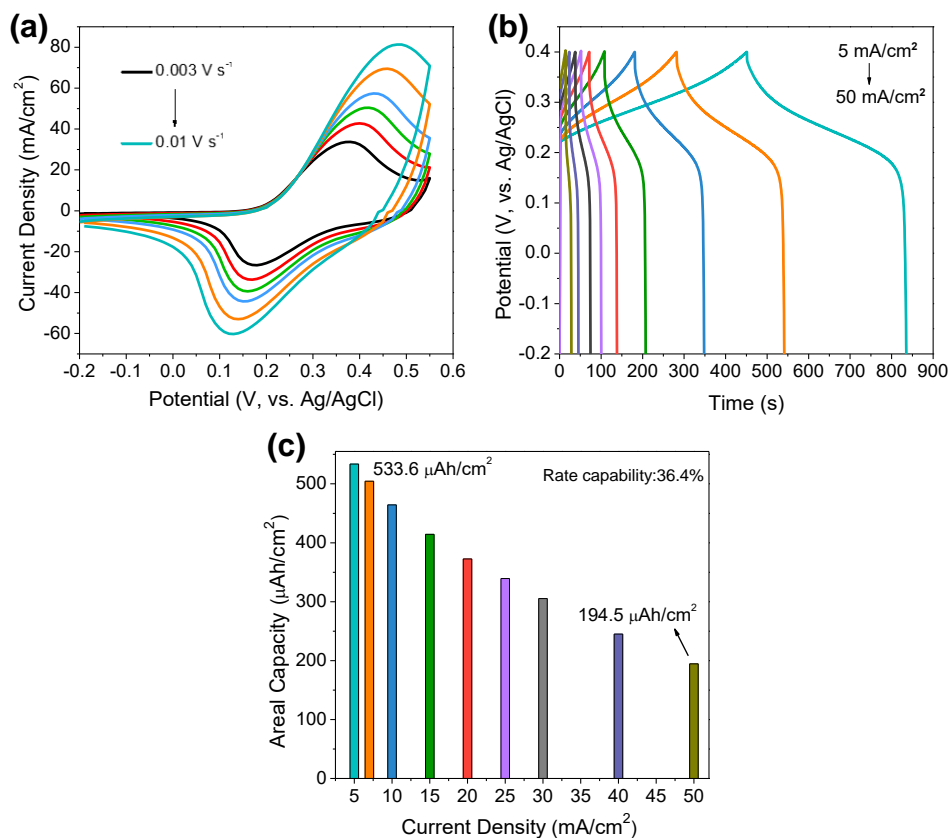


Fig. S3 Electrochemical properties of the monolayered MOF-based NCM-2 tested in 1 M KOH electrolyte. (a) CV curves evaluated at different scan rates of 0.003-0.01 V s⁻¹, (b) GCD curves measured at different current densities of 5-10 mA cm⁻² and (c) estimated capacity values as a function of current density for the monolayered MOF-based NCM-2

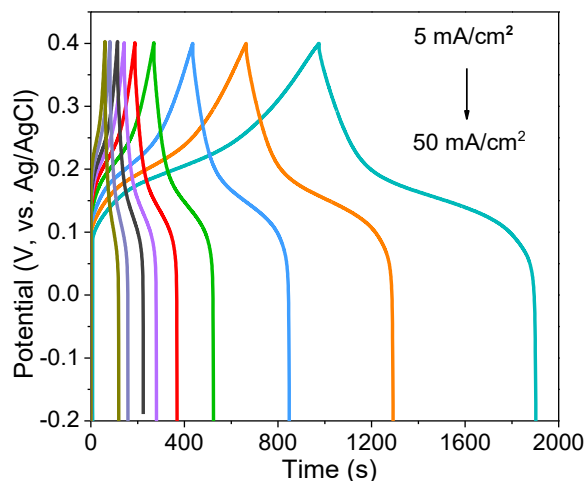


Fig. S4 GCD curves measured at different current densities of 5-10 mA cm⁻² for the dual layered MOF-based NCM-5

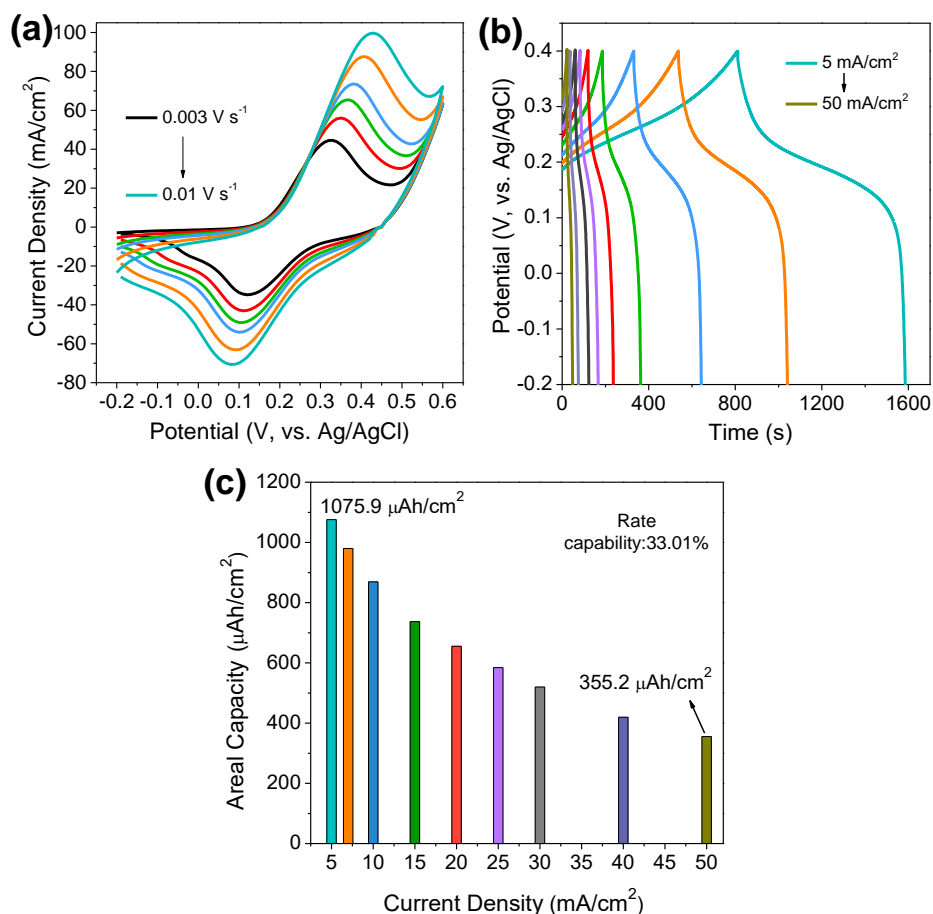


Fig. S5 Electrochemical properties of the MOF-based NCM-10 evaluated in 1 M KOH electrolyte. (a) CV curves evaluated at different scan rates of 0.003-0.01 V s⁻¹, (b) GCD curves measured at different current densities of 5-10 mA cm⁻² and (c) calculated capacity values with respect to the current densities of the MOF-based NCM-10

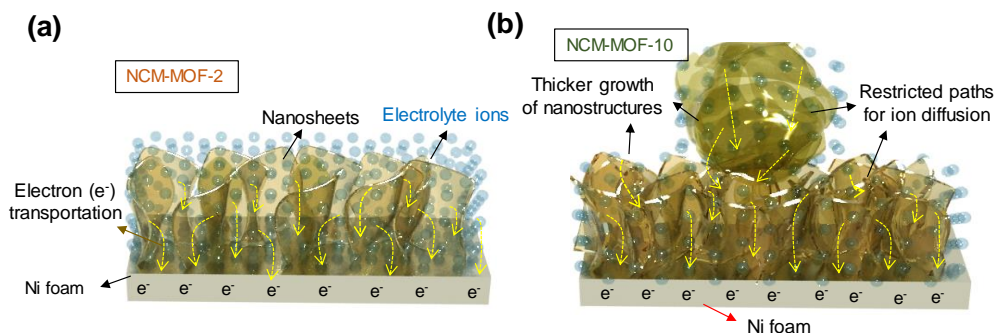


Fig. S6 Schematic prediction of electrochemical merits/dissontvages of (a) NCM-based MOF-2 and (b) NCM-based MOF-10 electrodes during the energy storage process

Table S1 Comparative energy storage performance of our dual layered NCM-based MOF with earlier reported metal oxide/sulfide and MOF-based materials in three-electrode system

Electroactive material	Morphology	Electrolyte	Test condition (mA cm ⁻²)	Area (cm ²)	Areal capacity (μAh cm ⁻²)	Refs.
MOF-74 derived Ni _x Co _{3-x} O ₄	Nanoparticles	6 M KOH	6	1	325.83	[S1]
Co-MOF	nanosheets	5 M KOH	4.5	1	1121.75	[S2]
Co-MOF derived Co ₉ S ₈ @S,N-doped carbon materials	Cuboid structure	6 M KOH	2.4	1	143	[S3]
Core-shell Ni-MOF-74@Co-MOF-74 derived Ni ₃ S ₄ @Co ₉ S ₈	Shell-in-shell tubes	2 M NaOH	2.4	1	367.47	[S4]
Ni-MOF	Accordion-like nanosheets	3 M KOH	3.5	1	638.12	[S5]
Ni-MOF/CNTs	Nanosheets	6 M NaOH	2	1	784.44	[S6]
Ni-MOF	Sheet-like structure	6 M KOH	2.5	1	547.84	[S7]
MOF-derived NiO	Porous nanoparticle	6 M KOH	5	1	201.25	[S8]
MOF-derived NiO/ZnO	Hierarchical hollow spheres	3 M KOH	5.2	1	276.11	[S9]
Zn/Co-ZIF derived Zn-Co-S	Rhombic dodecahedral cages	6 M NaOH	1	1	175.83	[S10]
Ni-based MOF	Nanorods	2 M KOH	1.9	1	131.1	[S11]
Bimetallic NiCo-MOF-74	Spherical particles	6 M KOH	3	1	238.33	[S12]
Ni-MOF derived NiO	Hexagonal flakes	3 M KOH	2	2	74.09	[S13]

MOF derived Ni–Co–S	Hierarchical nanosheets	1 M KOH	0.45	1	123.10	[S14]
MOF-derived hollow porous Ni _x P _y O _z	Microrods	2 M KOH	1.5	1	305.06	[S15]
NCM-based MOF	Dual layered nanosheets/nanoflowers	1 M KOH	5	1	1311.4	This work

Table S2 Comparative areal energy and power density values of recently reported hybrid devices with our dual layered NCM-based MOF//AC supercapattery

Positive electrode material	Negative electrode material	Potential window	Areal E _d (mWh cm ⁻²)	Areal P _d (mW cm ⁻²)	Refs.
Accordion-like Ni-MOF superstructure	Activated carbon	1.2 V	0.046	0.6	[S5]
Ni-MOF/CNTs	rGO/g-C ₃ N ₄	1.6 V	0.1648	1.8	[S6]
Onion-like nanoporous CuCo ₂ O ₄ hollow spheres	Activated carbon	1.5 V	0.2437	3.75	[S16]
MOF derived Ni–Co–S	Activated carbon	1.7 V	0.0955	3.27	[S14]
MOF derived MnO _x –MHCF nanocubes	Activated carbon	1.3 V	0.0410	0.32	[S17]
Hierarchical porous Ni-MOF	Activated carbon	1.4 V	0.0971	1.43	[S18]
Ni-based MOF	Activated carbon	1.2 V	0.2	1.8	[S19]
MOF derived Nanoporous Co ₃ O ₄	Nanoporous Carbon	1.6 V	0.0718	3.2	[S20]
ZnO QDs/carbon/CNTs	N-doped carbon/CNTs	1.7 V	0.1030	3.69	[S21]
Ni-Co MOF	Activated carbon		0.64	2.57	[S22]
Co-Mn MOFs	Activated carbon	1.6	0.12	4.68	[S23]
Dual layered NCM-based MOFs	Activated carbon	1.55 V	1.21	5.3	This work



Fig. S7 Powering multifunction electronic display by solar charge solitary supercapattery

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