

Supporting Information for

High-Index-Faceted Ni₃S₂ Branch Arrays as Bifunctional Electrocatalysts for Efficient Water Splitting

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Supplementary Figures

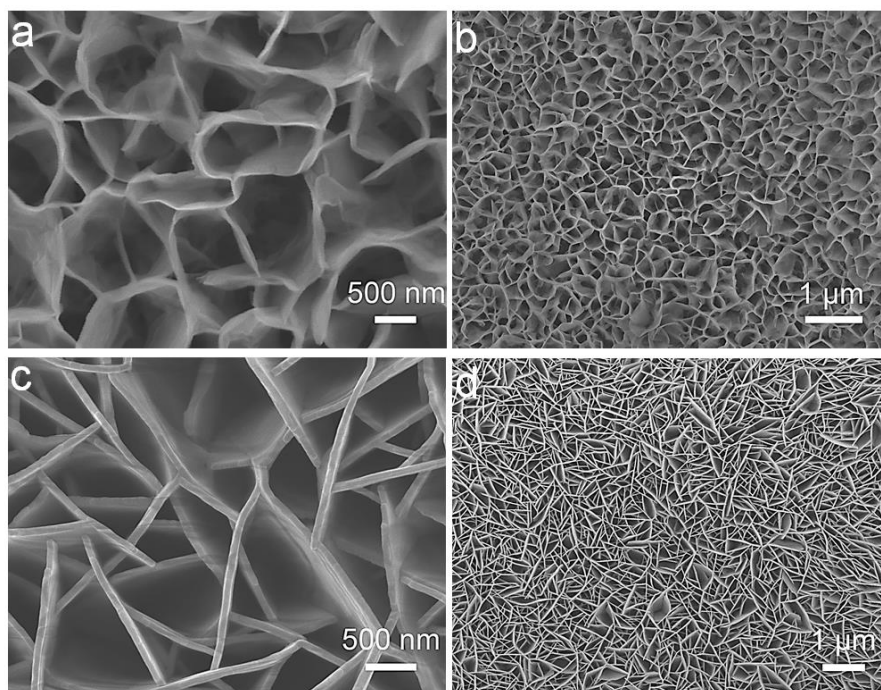


Fig. S1 SEM images of $\text{Ni}_2(\text{OH})_2\text{CO}_3$ and $\text{TiO}_2@ \text{Ni}_2(\text{OH})_2\text{CO}_3$ nanoflake arrays

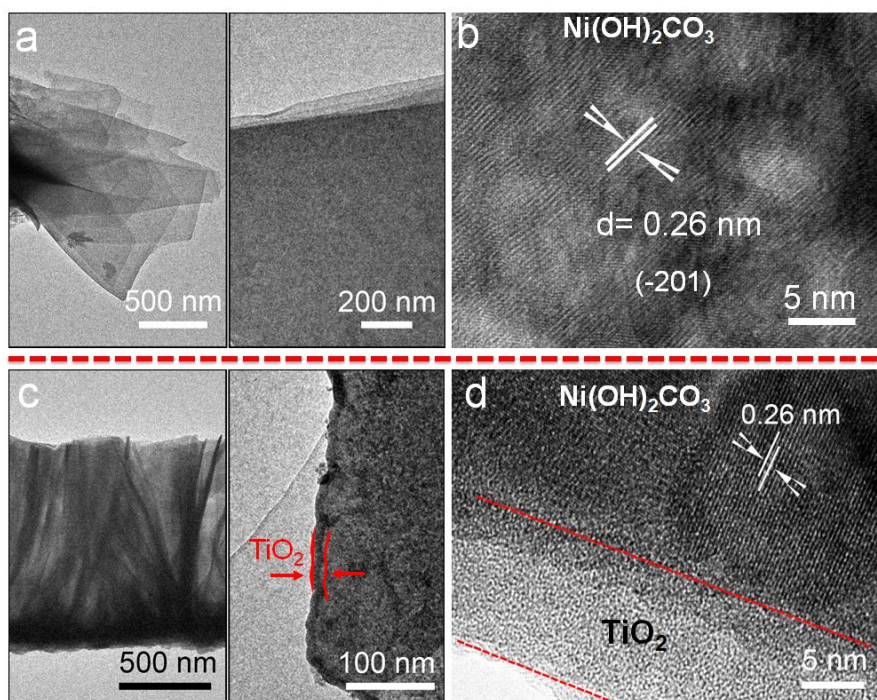


Fig. S2 TEM images of $\text{Ni}_2(\text{OH})_2\text{CO}_3$ and $\text{TiO}_2@ \text{Ni}_2(\text{OH})_2\text{CO}_3$ nanoflakes

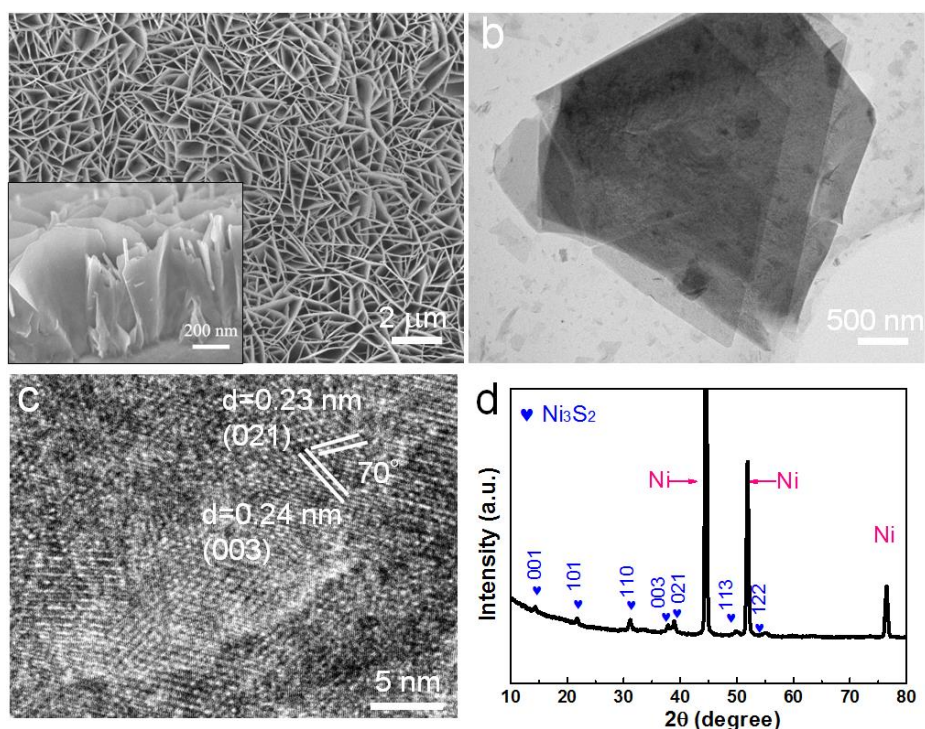


Fig. S3 Morphology and structure of control sample of pure Ni_3S_2 nanoflake arrays prepared without the TiO_2 support: **a** SEM image (fine structure in inset); **b, c** TEM-HRTEM images; **d** XRD pattern

SEM image (**Fig. S3a**) presents the uniform distribution of Ni_3S_2 nanoflakes on the nickel foam. TEM-HRTEM images (**Fig. S3b, c**) show the formation of regular Ni_3S_2 nanoflakes. When the HRTEM image is performed along the $[100]$ crystallographic direction of Ni_3S_2 , the (003) and (021) plane as well as the interior angle between the (003) and (021) facets of $\sim 70^\circ$ can be still detected, which is also corresponding with the exposed $\{\bar{2}10\}$ facet. XRD pattern further suggests the successful preparation of Ni_3S_2 (JCPDS 44-1418) (**Fig. S3d**).

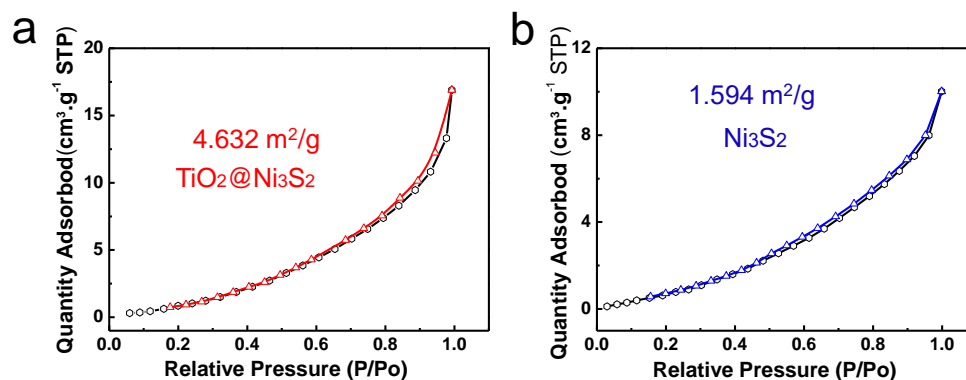


Fig. S4 BET measurements: nitrogen adsorption-desorption isotherm curves: **a** TiO_2 @ Ni_3S_2 nanowire arrays and **b** Ni_3S_2 arrays

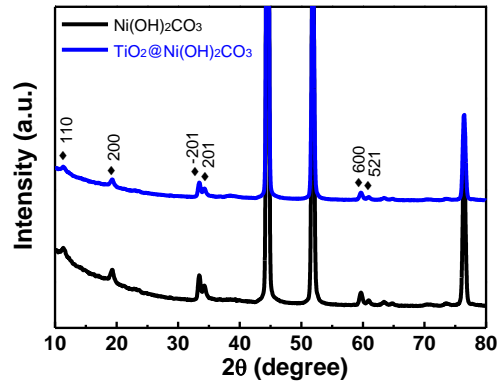


Fig. S5 XRD pattern of $\text{Ni}_2(\text{OH})_2\text{CO}_3$ and $\text{TiO}_2@\text{Ni}_2(\text{OH})_2\text{CO}_3$ arrays

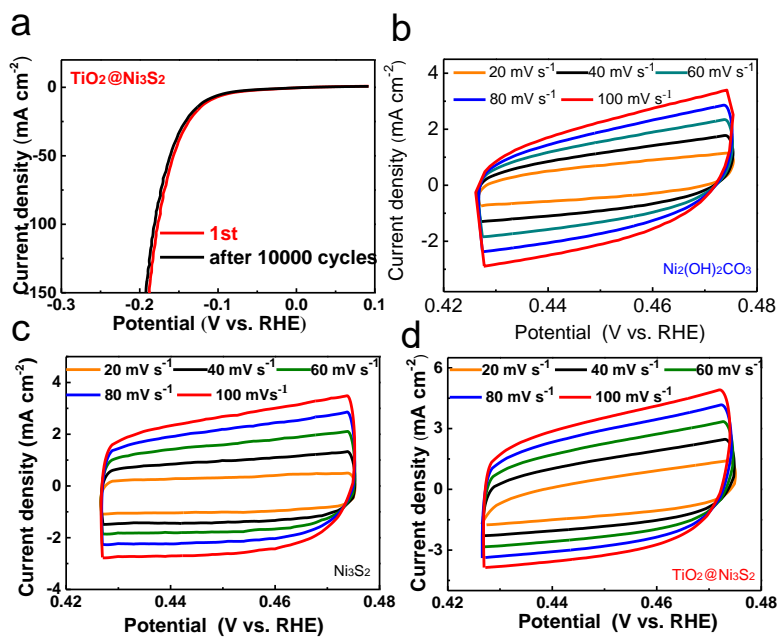


Fig. S6 **a** LSV plots for HER of the $\text{TiO}_2@\text{Ni}_3\text{S}_2$ electrode in the first and after 10,000 cycles. CV curves of different electrodes in double layer region at various scan rates: **b** $\text{Ni}_2(\text{OH})_2\text{CO}_3$, **c** Ni_3S_2 , and **d** $\text{TiO}_2@\text{Ni}_3\text{S}_2$ electrodes

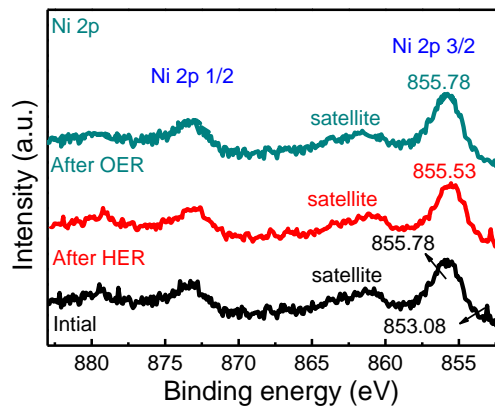


Fig. S7 XPS spectra of Ni 2p: **a** Initial state, **b** After HER and **c** After OER tests