

Supporting Information for

## Bioinspired MXene-Based Soft Actuators Exhibiting Angle-Independent Structural Color

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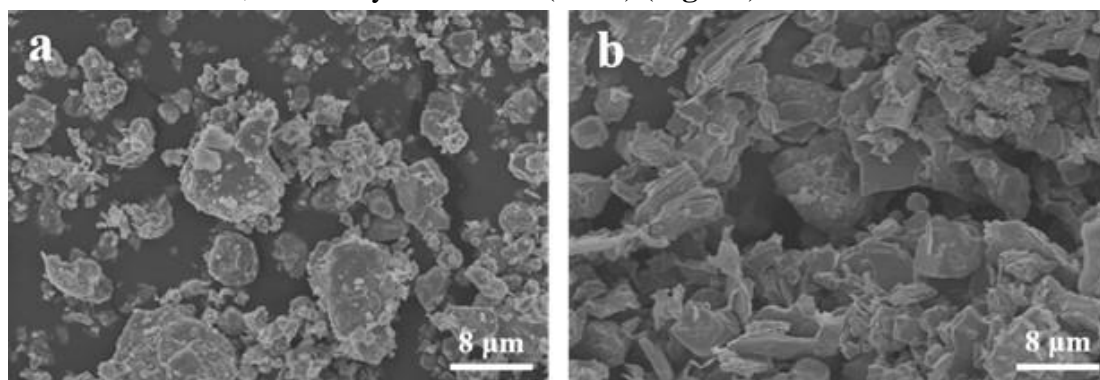
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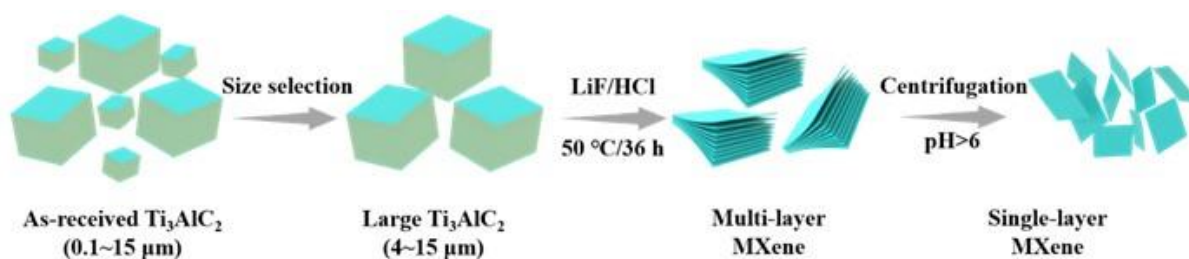
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### S1 Fabrication and Characterizations of MXene-based Soft Actuators

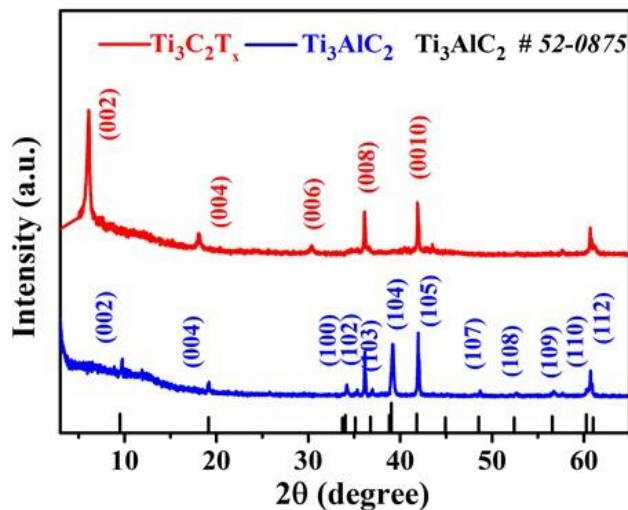
We synthesized MXene using LiF-HCl etching of  $\text{Ti}_3\text{AlC}_2$  MAX phase by preselecting large  $\text{Ti}_3\text{AlC}_2$  MAX phase particles from the as-received MAX phase prior to the selective etching step (**Fig. S1**), also referred to as the modified minimally intensive layer delamination (MILD) method, as illustrated in **Fig. S2**. All the X-ray diffraction (XRD) peaks can be well indexed to  $\text{Ti}_3\text{C}_2\text{T}_x$  (red line), which demonstrates that  $\text{Ti}_3\text{AlC}_2$  (JCPDS No. 52-0875, blue line) was successfully etched of Al layers (**Fig. S3**). The thin and transparent flakes of exfoliated  $\text{Ti}_3\text{C}_2\text{T}_x$  nanosheets can be clearly seen in transmission electron microscopy (TEM) image, showing the typical 2D sheet-like structure (**Fig. S4a**). The high-resolution images show that the constant interplanar distance is  $\sim 0.13$  nm and MXene exhibits a hexagonal crystalline nature, as shown by the selected area electron diffraction (SAED) pattern in **Fig. S4b**. These results are implemented to structural characterization of ultrathin MXene, indicating a successful formation of single-layer MXene nanosheets using a modified minimally intensive layer delamination (MILD) method. In addition, the dynamic light scattering characterization shows that the average hydrodynamic diameters of MXene nanosheets in *N,N*-dimethylformamide (DMF) are  $\sim 1$   $\mu\text{m}$  (**Fig. S5**). We also demonstrated that the lyotropic liquid crystalline phase of MXene nanosheets can be achieved using a solvent compatible with MXene, including deionized water and *N,N*-dimethylformamide (DMF) (**Fig. S6**).



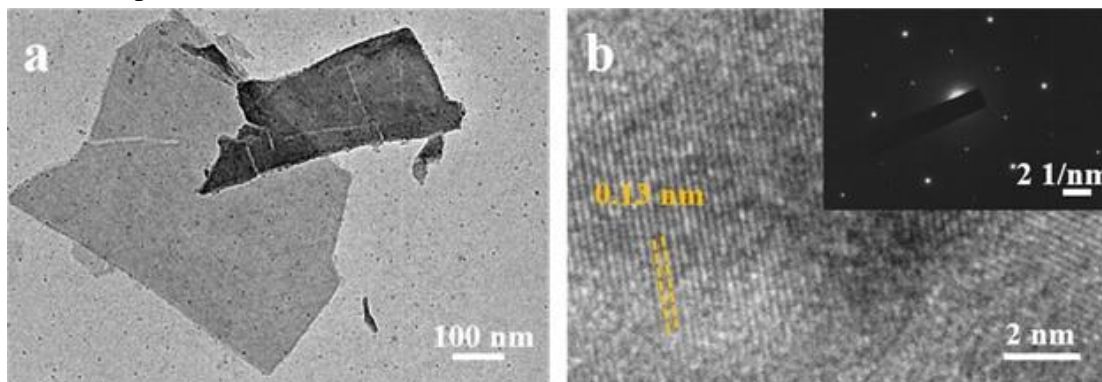
**Fig. S1** SEM images of  $\text{Ti}_3\text{AlC}_2$  (a) before and (b) after size selection



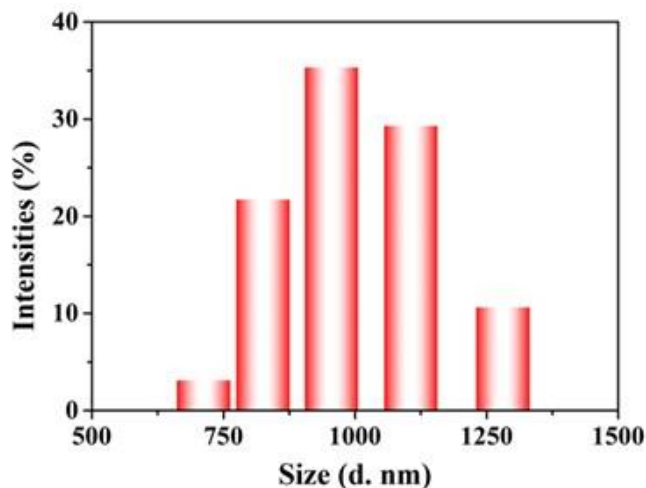
**Fig. S2** Schematic process of modified minimally intensive layer delamination (MILD) method



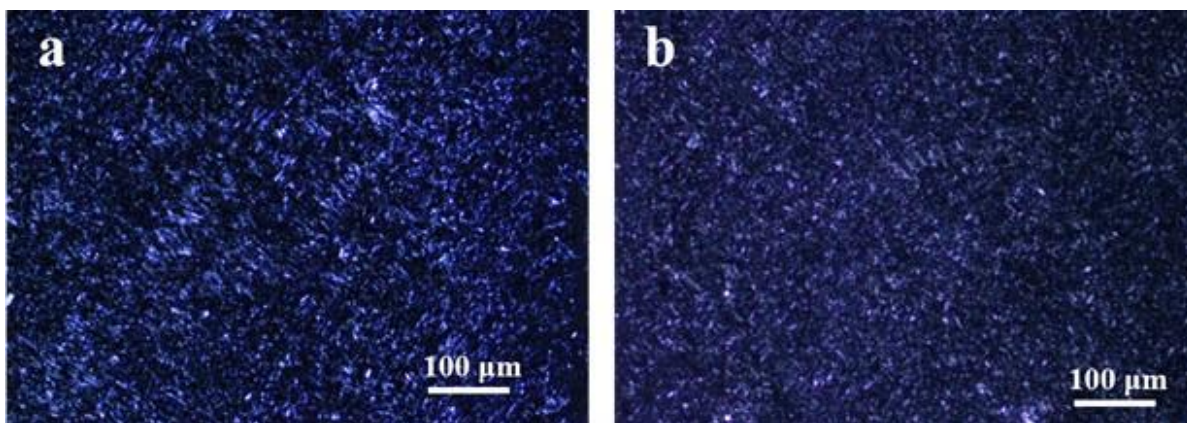
**Fig. S3** XRD patterns of the  $\text{Ti}_3\text{AlC}_2$  and MXene



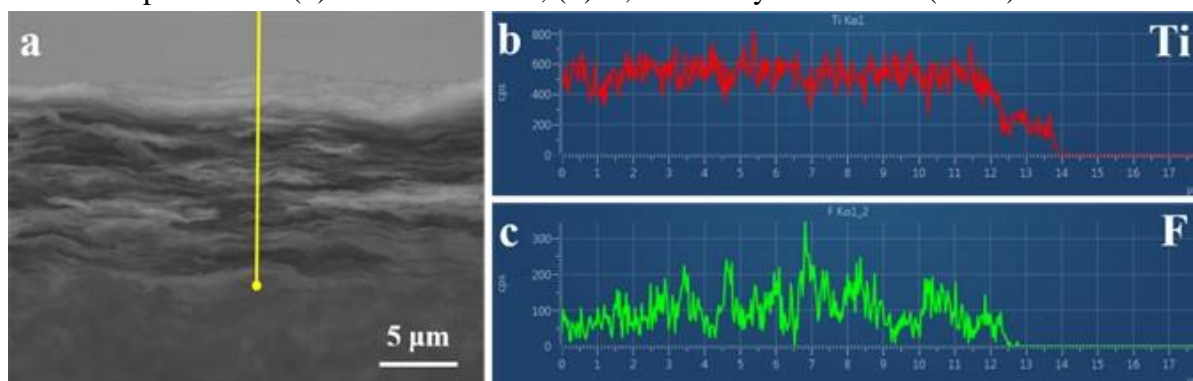
**Fig. S4** (a) TEM image of the single-layer MXene. (b) Magnified TEM image of the single-layer MXene, and the corresponding SAED pattern in the inset



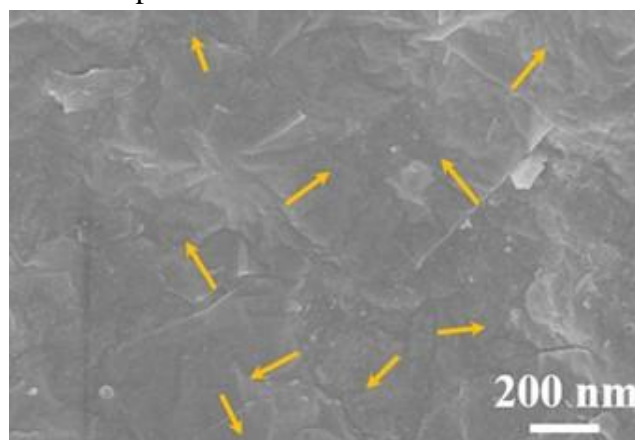
**Fig. S5** Particle-size distribution of MXene dispersions in N,N-dimethylformamide (DMF)



**Fig. S6** Characterizations of colloidal MXene liquid crystals: POM images of 28 mg/mL MXene dispersions in (a) deionized water, (b) *N,N*-dimethylformamide (DMF)



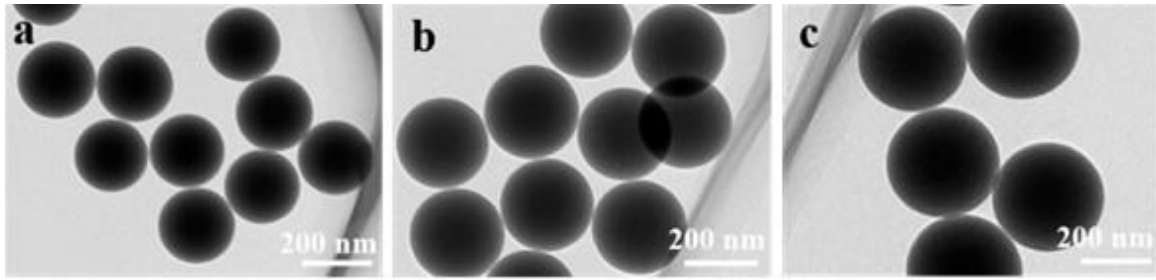
**Fig. S7** SEM image and EDX spectra of the MXene film



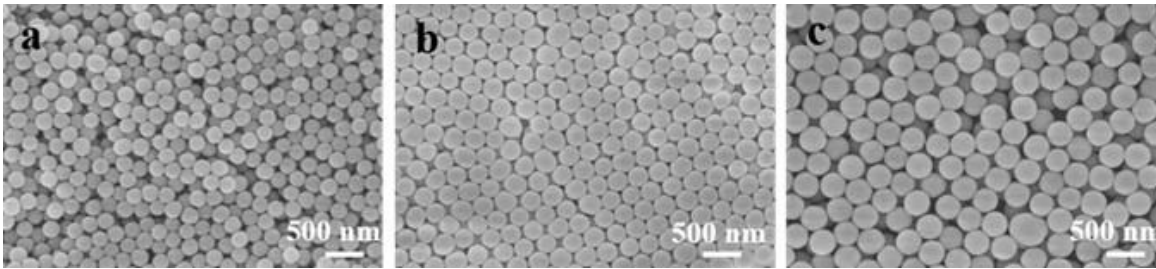
**Fig. S8** SEM image of the surface of the MXene film

## S2 Optical Properties of Angle-Independent Structural Color Soft Actuators

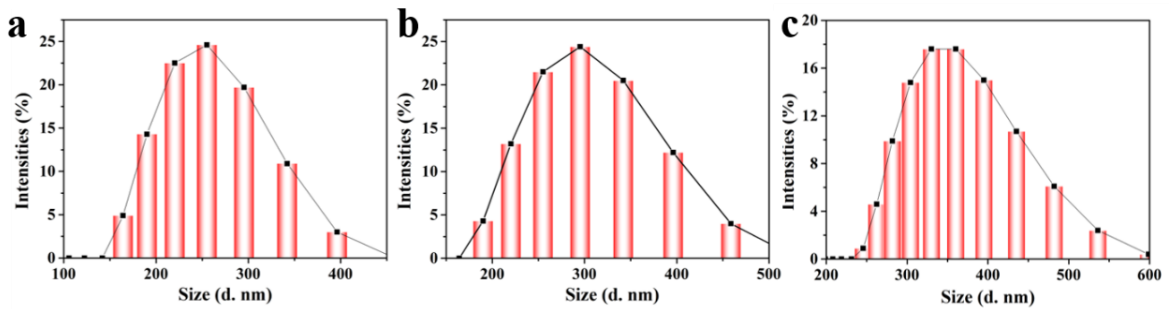
The preparation procedure of SiO<sub>2</sub> nanospheres was a modified Stöber process. TEM, SEM images taken from both the interior and the surface of samples show that the whole structure is quite homogeneous and the average diameters of SiO<sub>2</sub> nanospheres are ~210, 260, and 315 nm (**Figs. S9 and S10**). In addition, the dynamic light scattering (DLS) characterization shows that the average hydrodynamic diameters of SiO<sub>2</sub> nanospheres in deionized water are ~236 nm, 289 nm, 340 nm (**Fig. S11**). Hydrodynamic sizes of these microspheres measured by nanoparticle size measurement are bigger than the actual measured diameters from the SEM and TEM. DLS exhibits a hydrodynamic size that corresponds to the core and the swollen corona of the micelles, whereas SEM and TEM often exhibit the size of the core for micelles in a dried state as the corona with low electronic density is not visible [S1].



**Fig. S9** TEM images of different SiO<sub>2</sub> nanospheres: (a) 210 nm, (b) 260 nm, (c) 315 nm

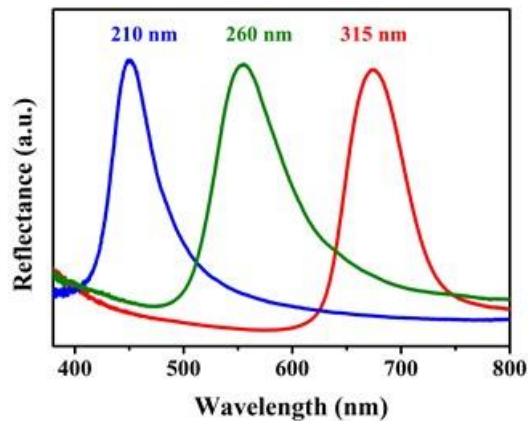


**Fig. S10** SEM images of different SiO<sub>2</sub> nanospheres: (a) 210 nm, (b) 260 nm, (c) 315 nm

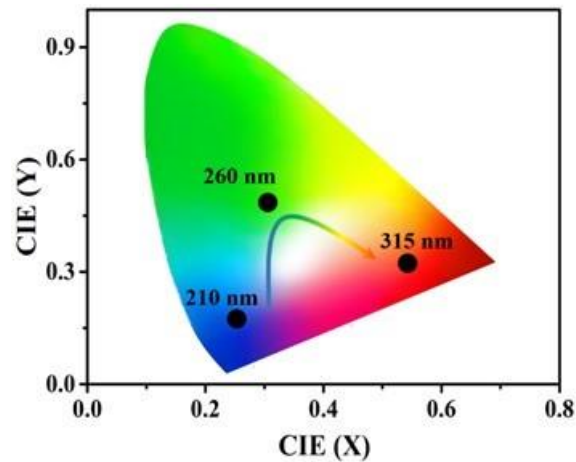


**Fig. S11** Particle-size distribution of different SiO<sub>2</sub> nanospheres: (a) 210 nm, (b) 260 nm, (c) 315 nm dispersions in ethyl alcohol

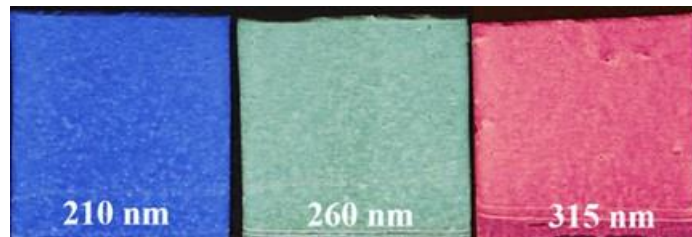
The MXene/SiO<sub>2</sub>@PVDF films with SiO<sub>2</sub> nanospheres of 210, 260, and 315 nm in diameter showed reflection peaks at 445, 550, and 662 nm, respectively (**Fig. S12**). The measured reflection spectra of these films were converted into the more standardized Commission Internationale de L'Eclairage (CIE) chromaticity value, which verified that the blue, green, red-structural color films were prepared (**Fig. S13**). **Figure S14** presents digital images of three different colored films, blue, green and red, which are composed of SiO<sub>2</sub> nanospheres with diameters of 210, 260, and 315 nm, respectively. **Figure S15** shows the barely changes of the reflection spectra peak positions of the three colored MXene/SiO<sub>2</sub>@PVDF film as the incidence angles increased from 0° to 60°.



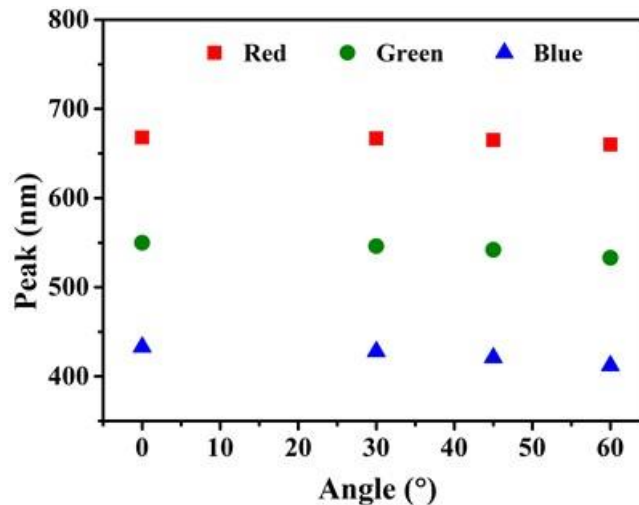
**Fig. S12** The reflection spectra of the three colored MXene/SiO<sub>2</sub>@PVDF films with 210, 260, and 315 nm diameter sizes



**Fig. S13** Positions of the three colored MXene/SiO<sub>2</sub>@PVDF films in the Commission Internationale de L'Eclairage (CIE) chromaticity diagram

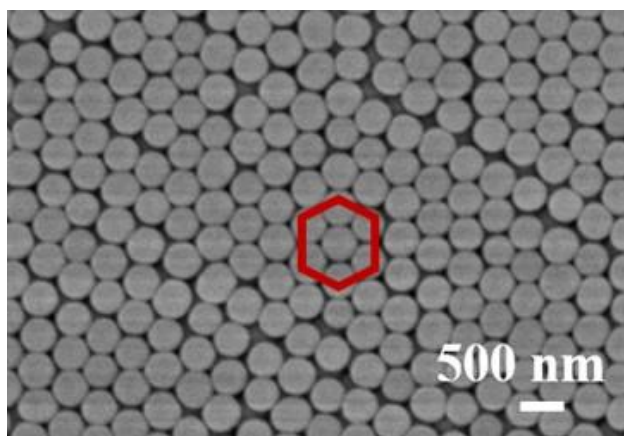


**Fig. S14** Digital photographs of the three colored MXene/SiO<sub>2</sub>@PVDF films, blue (210 nm SiO<sub>2</sub> nanospheres), green (260 nm SiO<sub>2</sub> nanospheres), and red (315 nm SiO<sub>2</sub> nanospheres)

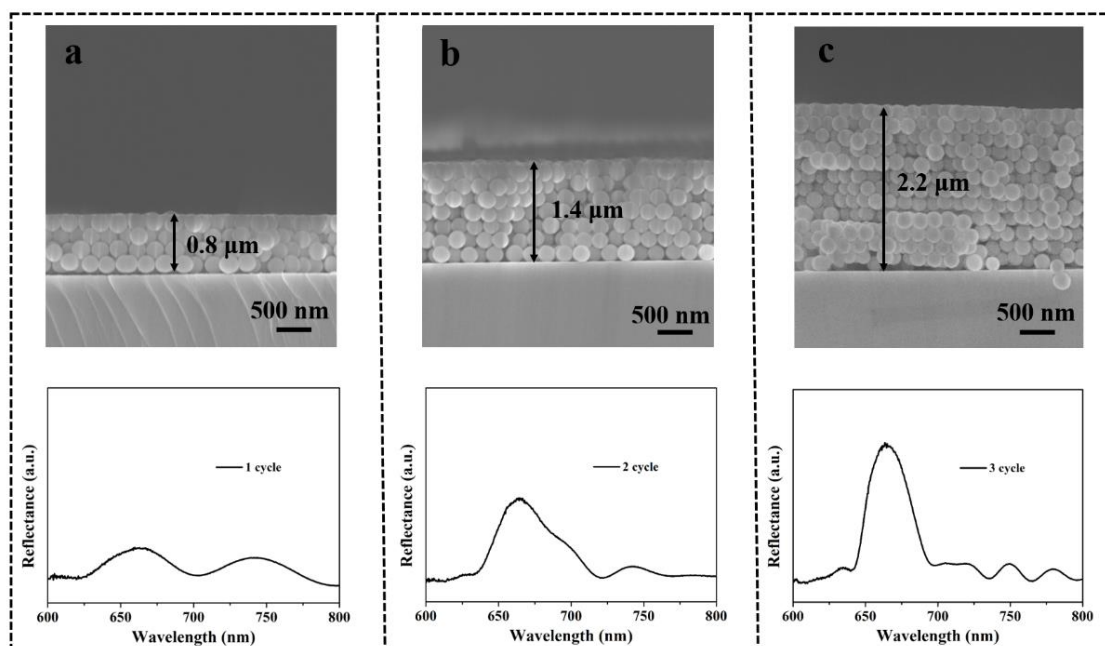


**Fig. S15** Changes of the reflection spectra peak positions of the three colored MXene/SiO<sub>2</sub>@PVDF films as the incidence angles increased from 0° to 60°

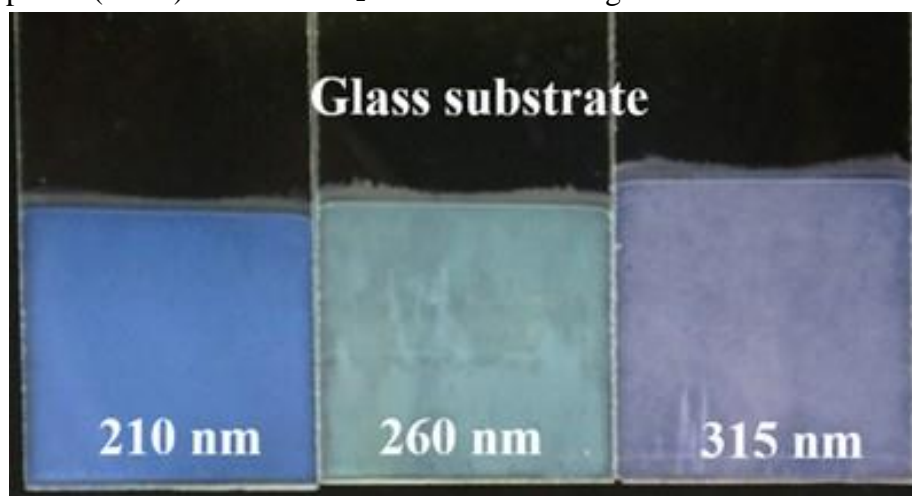
In contrast, the 3D photonic crystals with only SiO<sub>2</sub> nanoparticles, in which high long-range order was obtained by directly dip-coating colloidal SiO<sub>2</sub> nanoparticles on the glass substrate, are named SiO<sub>2</sub> photonic crystal films. The SiO<sub>2</sub> nanospheres were assembled into highly ordered SiO<sub>2</sub> photonic crystal films, as marked by a red hexagon (FCC ordered arrays), which can be observed on the top-surface and cross-sectional scanning electron microscopy (SEM) images in **Figs. S16 and S17**. The highest reflection peak of the SiO<sub>2</sub> photonic crystal films was achieved when the dip-coating was three times, whereas the effect of the SiO<sub>2</sub> photonic crystal films thickness on the reflection spectra is also illustrated in **Fig. S17**. The digital photos of the SiO<sub>2</sub> photonic crystal films (210, 260, and 315 nm) exhibit different colors of blue, green, and red at a vertical angle (**Fig. S18**). Meanwhile, the SiO<sub>2</sub> photonic crystal films of diameters 210, 260, and 315 nm showed reflection peaks at 445, 550, and 662 nm, respectively (**Fig. S19**).



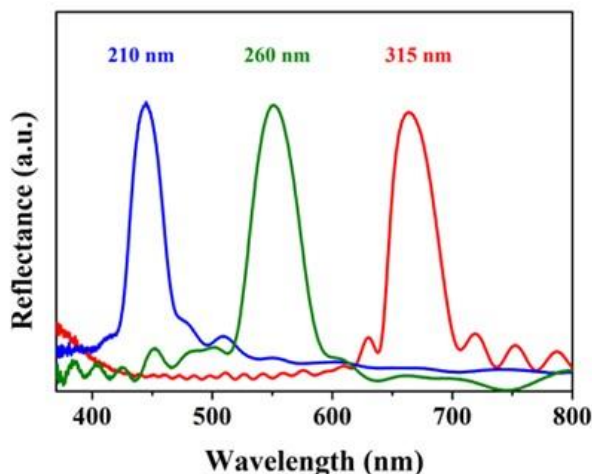
**Fig. S16** SEM image of the SiO<sub>2</sub> photonic crystal films



**Fig. S17** The SiO<sub>2</sub> photonic crystal films. The SiO<sub>2</sub> photonic crystal films assembled by dip-coating method for (a) one, (b) two, and (c) three times coating (top) and the corresponding reflection spectra (down). 315 nm SiO<sub>2</sub> was used in this figure

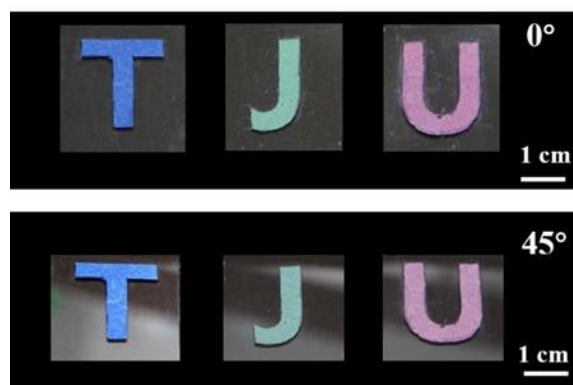


**Fig. S18** SiO<sub>2</sub> photonic crystal films assembled by three times dip-coating method on the glass substrate with 210, 260, and 315 nm diameter size

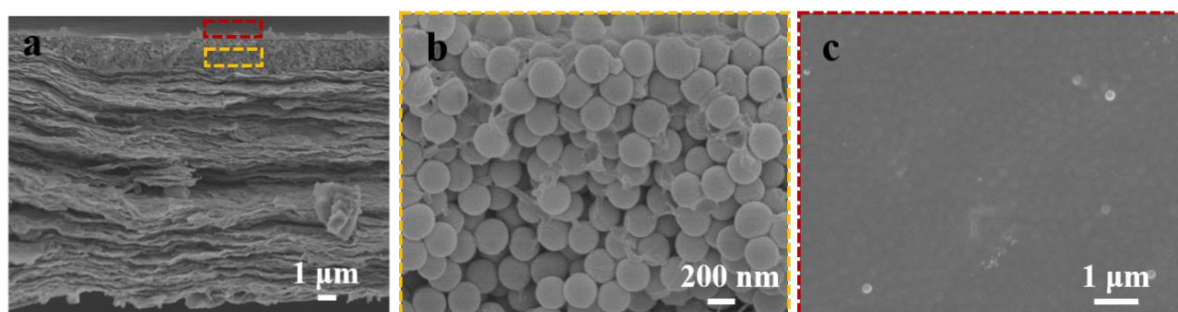


**Fig. S19** The reflection spectra of SiO<sub>2</sub> photonic crystal films on the glass substrate with 210 nm, 260 nm, and 315 nm diameter sizes

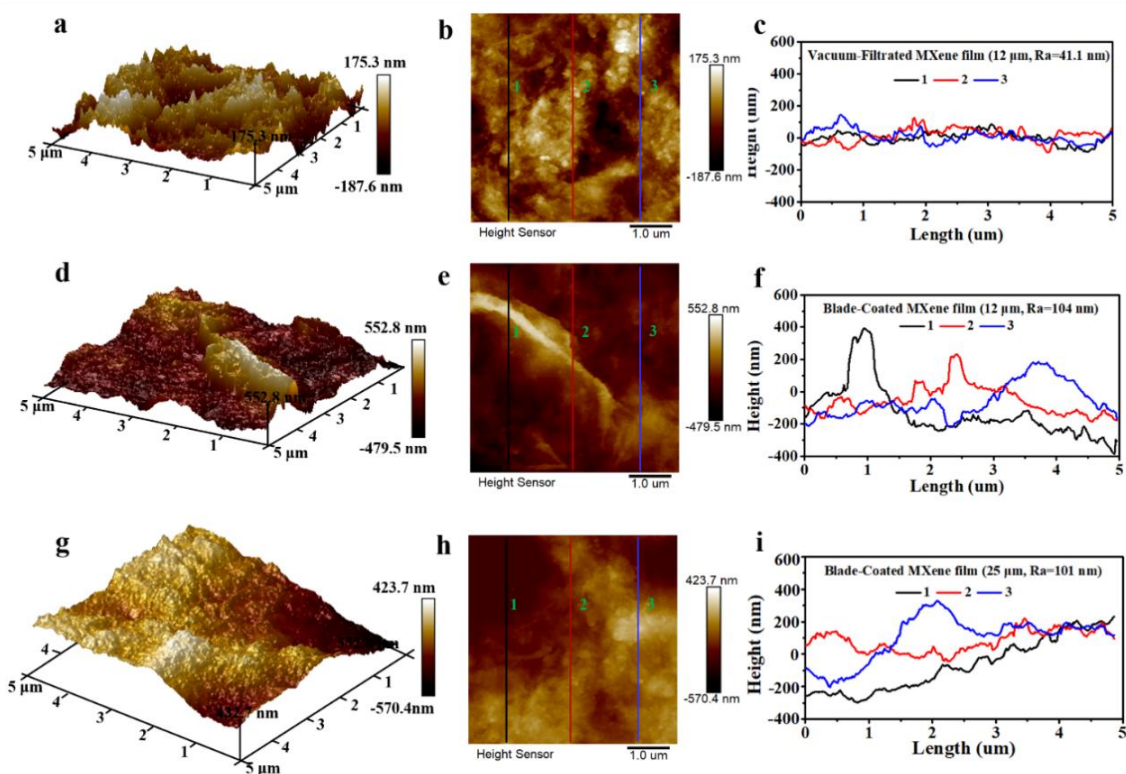
Photographs of different MXene/SiO<sub>2</sub>@PVDF films further indicate the appearance of the structural colors remained almost constant at viewing angles from 0° to 60° relative to the normal direction of the films (**Fig. S20**). The angle-independence of MXene-based structural color soft actuators was closely related to the formation of short-range ordered 3D amorphous photonic nanostructures, as shown in the cross-sectional TEM image of MXene/SiO<sub>2</sub>@PVDF film (**Fig. S21**). Taking advantage of controllable self-assembly with a dip-coating method, a MXene/SiO<sub>2</sub>@PVDF film with color gradient was prepared by regional dip-coating method. Firstly, a piece of glass slide was put in emulsions A, and dip-coated a certain height. The part which was immersed in emulsion A is cleaned up, and then the glass was put in emulsion B for the dip-coating process. Repeat the above steps for emulsion C. Finally, a tricolor template is obtained, which was made up of SiO<sub>2</sub> nanospheres with various diameters (**Fig. S24**).



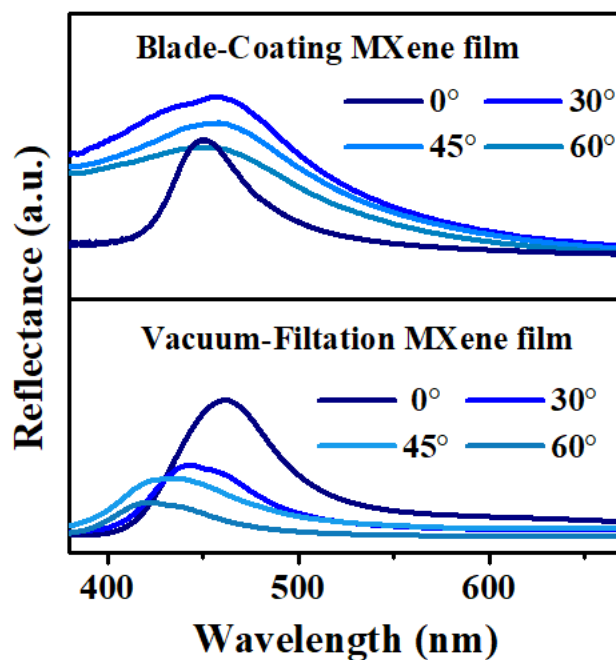
**Fig. S20** Photographs of the MXene/SiO<sub>2</sub>@PVDF films with “TJU” pattern taken at 0° and 45° incidence angles



**Fig. S21** Morphologies and characterizations of the MXene/SiO<sub>2</sub>@PVDF films. (a) The cross-sectional SEM image of MXene/SiO<sub>2</sub>@PVDF film. (b) The high-magnification cross-sectional SEM image of the SiO<sub>2</sub> layer. (c) The top-view SEM image of the SiO<sub>2</sub> layer

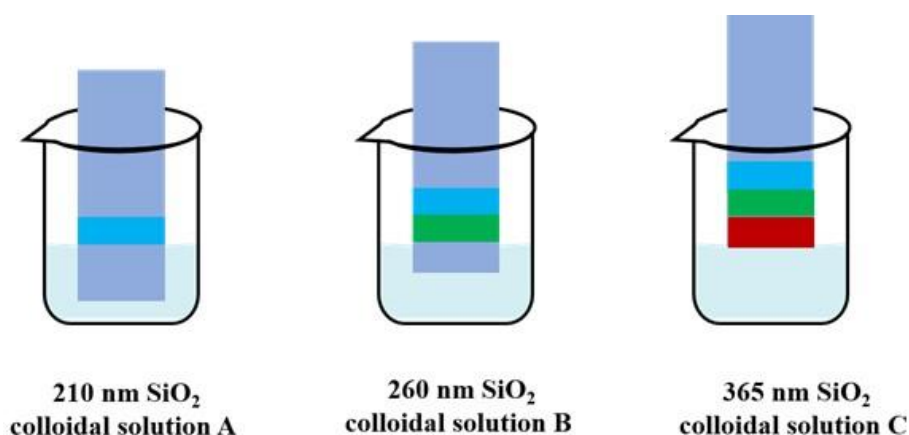


**Fig. S22** 3D and 2D AFM images of MXene film and height profiles. (a-c) 12  $\mu\text{m}$  MXene film prepared *via* the vacuum-filtration method. (d-f) 12  $\mu\text{m}$  MXene film prepared *via* the blade-coating method. (g-i) 25  $\mu\text{m}$  MXene film prepared *via* the blade-coating method



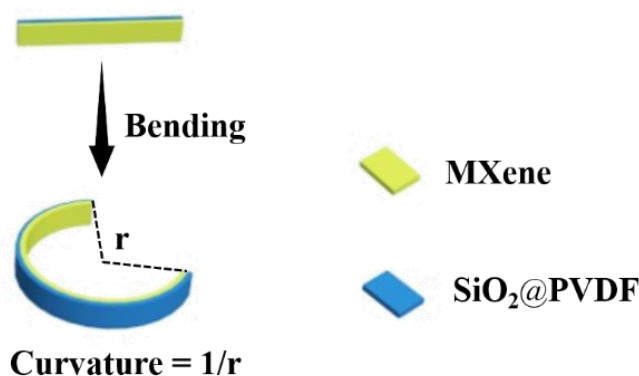
**Fig. S23** Reflection spectra of MXene/SiO<sub>2</sub>@PVDF films made by blade-coating MXene film and vacuum-filtration MXene film



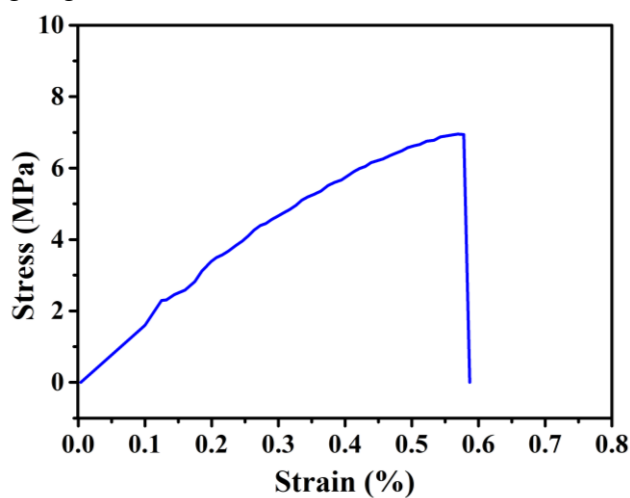


**Fig. S24** Schematic diagram of the regional dip-coating method

### S3 Actuation Performance of Angle-Independent Structural Color Soft Actuators



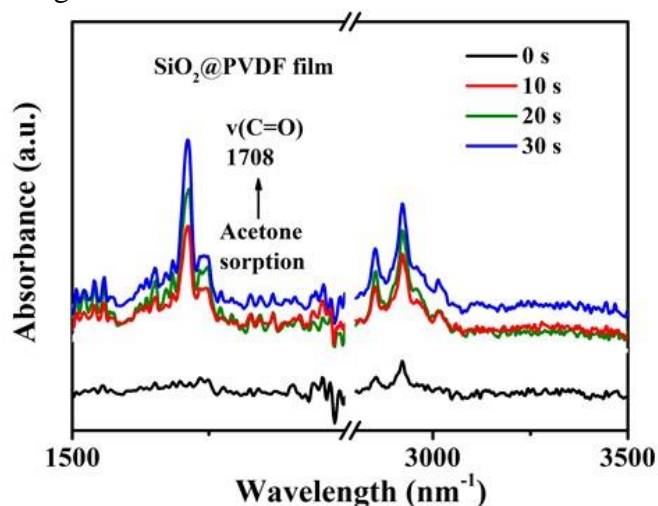
**Fig. S25** Schematic illustration of bending angle (curvature) in acetone vapor-driven structurally colored soft actuators. First, the structurally colored soft actuators are cut into strip ( $20 \text{ mm} \times 2 \text{ mm} \times 14 \mu\text{m}$ ). Second, the strip is attached to a pair of pincers. Then, the actuation is tested when the structurally colored soft actuators exposure to acetone vapor and an air environment. Finally, the  $1/r$  is the curvature of the structurally colored soft actuators to represent shape-bending angle



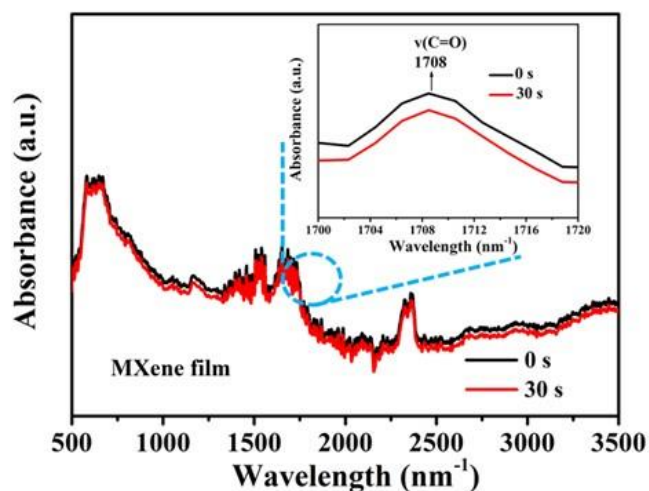
**Fig. S26** Strain-stress curve of the MXene/SiO<sub>2</sub>@PVDF film

The actuation performance of structural color soft actuators was then investigated under an acetone vapor environment. After exposure of SiO<sub>2</sub>@PVDF film to acetone vapor, a typical peak signal came out at  $1708 \text{ cm}^{-1}$ , which was enhanced with prolonging acetone vapor-

exposure time, indicating the adsorption of acetone molecules into PVDF structure (**Fig. S27**). **Figure S28** show the FT-IR spectra for monitoring the acetone vapor absorption of MXene film. The profiles of MXene film have no visible absorbance change at  $1708\text{ cm}^{-1}$  upon prolonging acetone vapor-exposure time. It is indicated that the MXene film shows low acetone adsorption due to its highly aligned and dense topography. In addition, a microbalance (Mettler Toledo) was utilized to monitor the mass variation of the films (**Fig. S29**). For the  $\text{SiO}_2@\text{PVDF}$  film, the initial weight was  $17.55\text{ mg}$  which was defined as  $W_0$ . After absorption of acetone vapor for  $10\text{ s}$ , the film's weight was recorded and marked as  $W_a$ . The film was then allowed to desorb acetone molecules in air for  $10\text{ s}$ , and the weight was recorded and defined as  $W_b$ . Total five cycles were repeated. Acetone vapor sorption ratio in each cycle was calculated as:  $(W_a - W_0) \times 100\% / W_0$  and  $(W_b - W_0) \times 100\% / W_0$  respectively. In the case of MXene film, the data was obtained as the same calculating method, and the initial weight of MXene was  $15.91\text{ mg}$ . The results indicate that  $\text{SiO}_2@\text{PVDF}$  film is capable of adsorption of acetone vapors, and desorbing them rapidly, but MXene film doesn't show significant mass change with acetone adsorption and desorption. Thus, the actuation mechanism of MXene/ $\text{SiO}_2@\text{PVDF}$  film is driven by the rapid and strong adsorption of acetone molecules in the hydrophobic fluorine groups of  $\text{SiO}_2@\text{PVDF}$  film, whereas the MXene layer shows low acetone adsorption and limited volume change due to its highly aligned and dense topography, thus yielding an asymmetric shape-bending deformation towards the MXene side.



**Fig. S27** FT-IR spectra for monitoring the acetone vapor absorption of  $\text{SiO}_2@\text{PVDF}$  film



**Fig. S28** FT-IR spectra for monitoring the acetone vapor absorption of MXene film

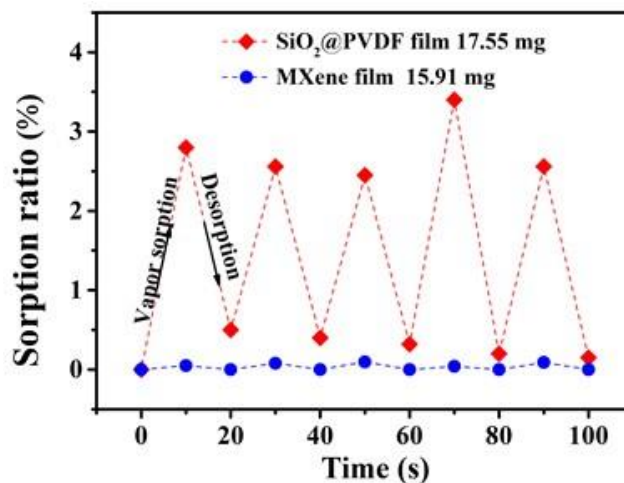


Fig. S29 Acetone-vapor absorption performance of SiO<sub>2</sub>@PVDF and MXene film

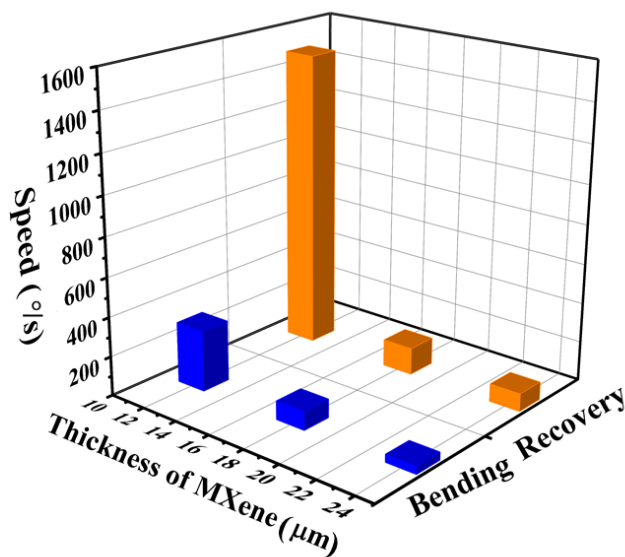


Fig. S30 Relationship between the thickness of MXene layer and bending/recovery speed

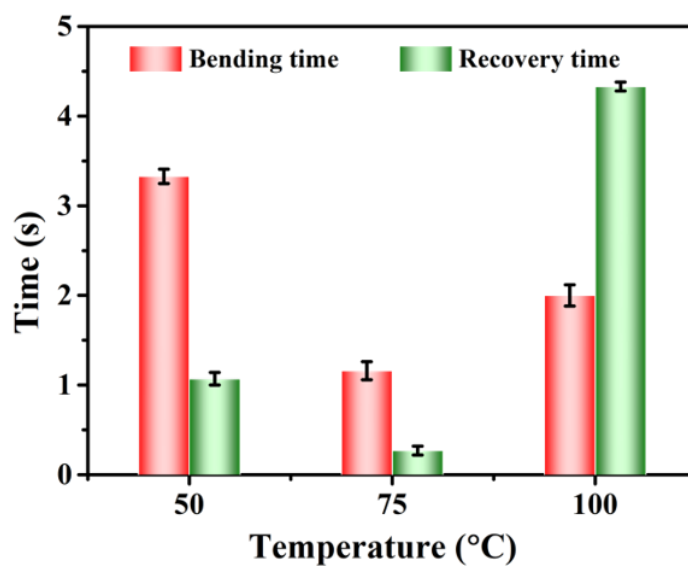
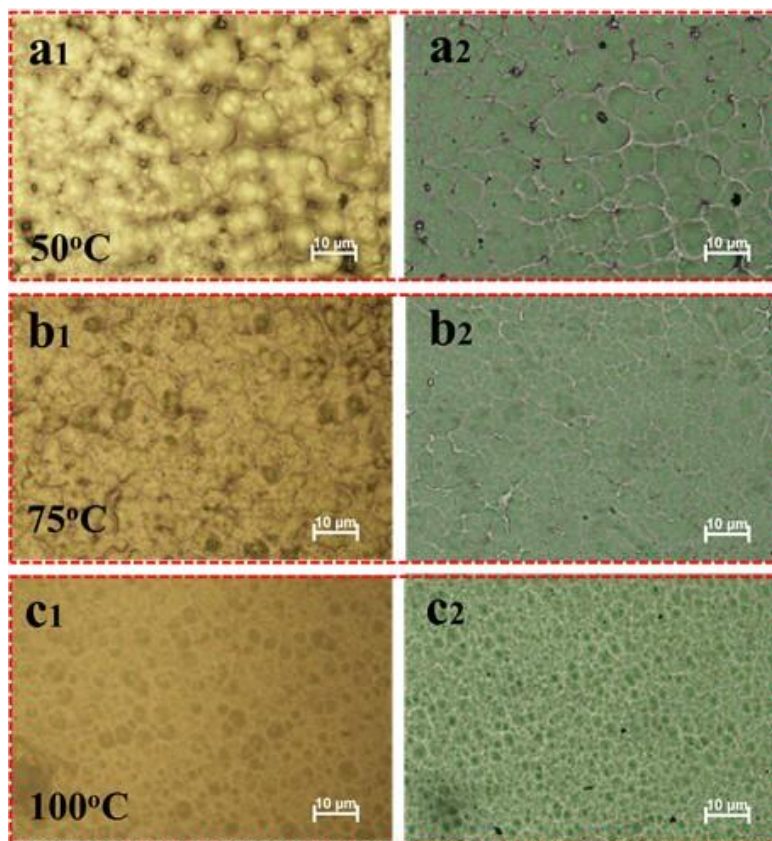


Fig. S31 Relationship between the solvent evaporation temperature of PVDF solution during the fabrication process and bending/recovery time

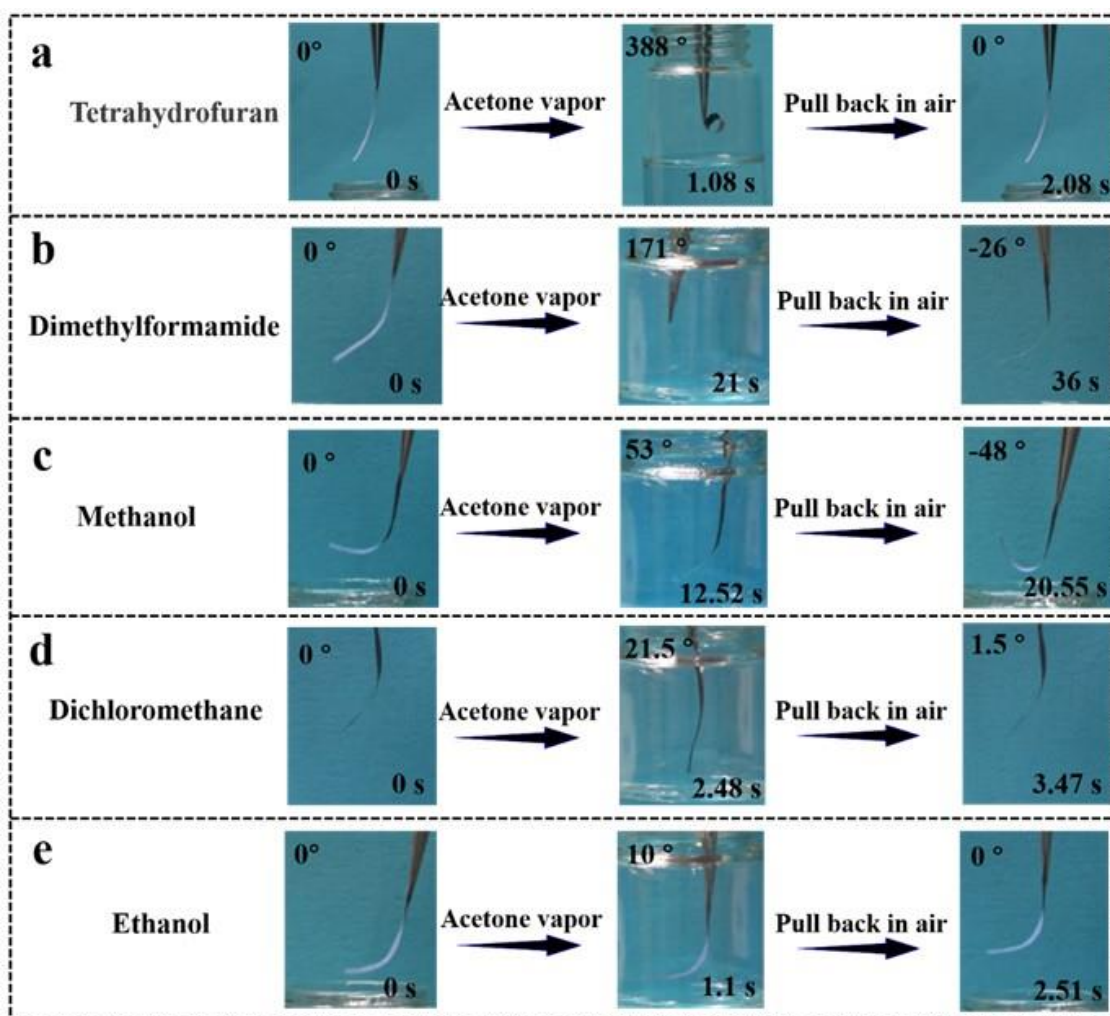


**Fig. S32** Optical microscope images of PVDF film deposited at different temperatures of solvent evaporation. **(a)** 50 °C, **(b)** 75 °C, and **(c)** 100 °C. **(a1-c1)** Using optical microscope images of the reflected light, **(a2-c2)** Using optical microscope images of the transmitted light.

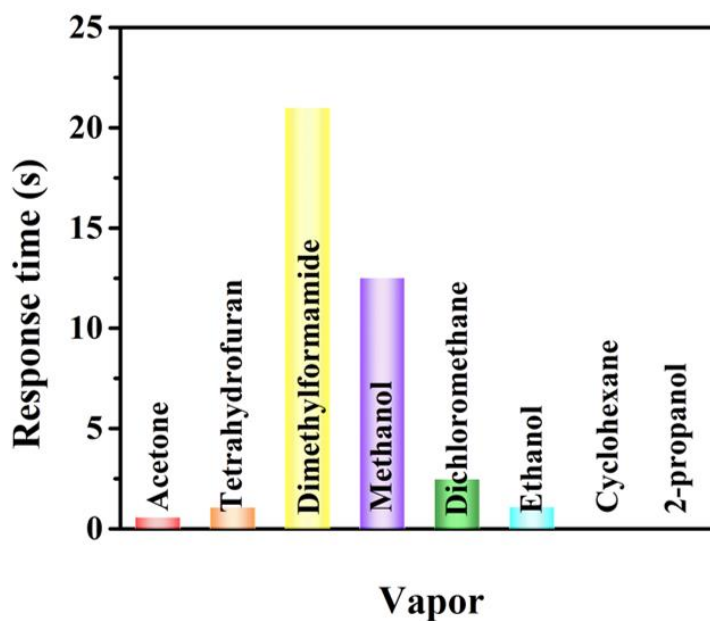
Optical microscope images of PVDF film deposited at different temperatures of solvent evaporation: 50, 75, and 100 °C. The diameter of formed microspheres at 100 °C reaches  $\sim 3 \mu\text{m}$  that results in a surface pore size much smaller and denser than that of the PVDF film prepared at 50 and 75 °C. Too small and dense pore size actually is favorable to acetone absorption but not favorable to facile desorption of acetone. Therefore, the soft actuators prepared at 100 °C of PVDF solvent evaporation could have the maximum bending curvature and slowly recovery speed

#### **S4 Biomimetic Proof-of-concept Illustrations Based on Angle-independent Structural Color Soft Actuators**

Upon exposure to acetone vapor, the swelling of the active layer causes downward bending of the initially curved, angle-independent structural color soft actuator. At this point, because the ratcheted substrate surface prevents backward sliding, the back edge bearing a higher frictional force ( $F_2 > F_1$ ) acts as a stationary end, leading to the front part of the soft actuator moving forward. Once the acetone vapor is removed, the deswelling process provides an unbending force (upward bending) that releases the soft actuator back to its original shape. In this situation, the front edge grips the ratcheted substrate and bears a higher frictional force ( $F_1' > F_2'$ ) accordingly, thus acting as a stationary end that allows forward movement of the back part. Once the recovery motion stops, the actuator is allowed to initiate a new downward and upward bending process after a new acetone absorption/desorption cycle, resulting in a walking soft robot that moves forward a longer distance along the direction of  $v$ , as shown in **Fig. S36**.



**Fig. S33** Photographs of the bending and recovering process of the angle-independent structural color soft actuators in the vapor of different types of solvents



**Fig. S34** Dependence of response time of the soft actuator on various organic vapors

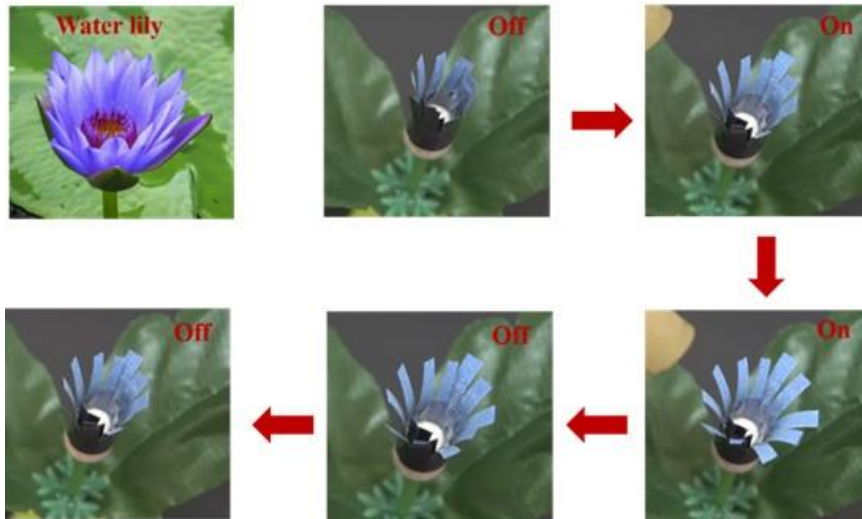


Fig. S35 The process of biomimetic flower in various stages of blooming

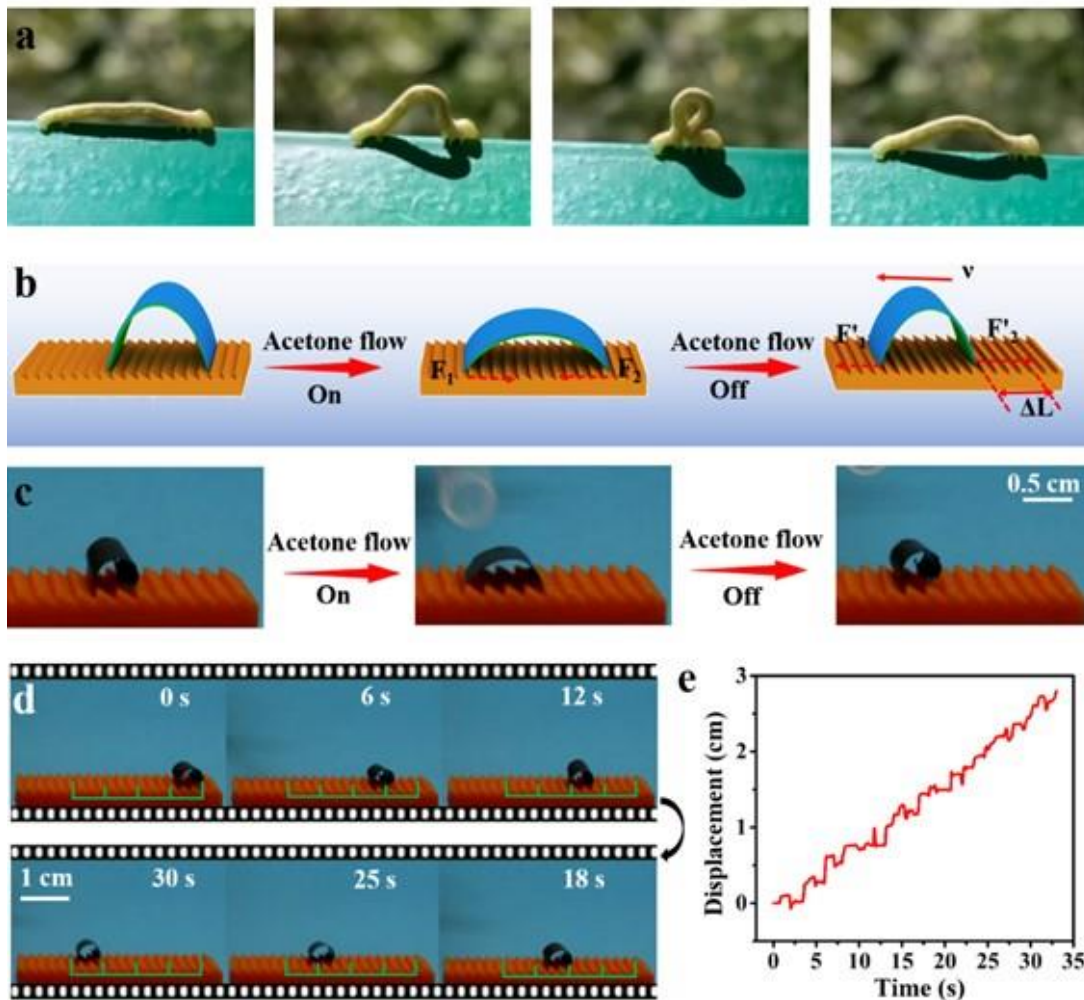


Fig. S36 Biomimetic walking soft robot. (a) The movement process of the inchworm in nature. (b, c) The schematic diagram of the walking soft robot and its movement. (d) The movement process of the walking soft robot under acetone vapor absorption/desorption cycles. (e) Horizontal displacement and velocity change of the walking soft robot along the horizontal position over time

**Table S1** Summary of different materials used for the design of actuators. Note: In each cited paper, we choose the most rapid speed for comparison with our results

Material	Response type	Response/Recovery time (s)	Color type	Curvature/Bending angle (mm <sup>-1</sup> )/(°)	Refs.
MXene/SiO <sub>2</sub> @PVDF	Acetone	1.16/0.24	Non-iridescent structural colors	0.52/360	This work
PNIPAM-K6APA	Temperature	8/211	Fluorescent colors	---/180	Angew. Chem. Int. Ed. <b>131</b> , 16243 (2019)
GO-PNIPAM	Temperature	600/---	Fluorescent colors	---/360	Adv. Funct. Mater. <b>28</b> , 1704568 (2018)
PNIPAM/rGO	NIR	120/---	Iridescent structural colors	---/60	Adv. Mater. <b>29</b> , 1704569 (2017).
LCN/GO	Visible light	2.94/3.53	Iridescent structural colors	---/97	Adv. Opt. Mater. <b>6</b> , 1800131 (2018).
PA/LCN	Humidity	10/---	Fluorescent colors	---/275	Adv. Funct. Mater. <b>31</b> , 2010578 (2021).
P(IL-co-MMA)	Humidity	4/---	Iridescent structural colors	2.1/1440	ACS Nano <b>12</b> , 12149 (2018).
PTMPTA	Acetone	---	Iridescent structural colors	---	Matter <b>1</b> , 626-638 (2019).
PEGDA-PUA	Acetone	15/85	Iridescent structural colors	---/90	J. Mater. Sci. <b>55</b> , 817-827 (2019).
PVDF	Acetone	12/---	---	3/---	Adv. Mater. <b>29</b> , 1702231 (2017).
PVDF/MS	Acetone	20/---	---	1/---	Mater. Eng. <b>304</b> , 1800501 (2019).
GO/CA	Acetone	19/38	---	---/720	Smart Mater. Struct. <b>28</b> , 105043 (2019).
cPVA/PVDF/CNT@PDMS	Acetone	5/25	---	0.65/---	Chem. Eng. J. <b>394</b> , 124919 (2020).
PCAD@AG	Humidity	7/33	---	---/260	Nat. Commun. <b>6</b> , 7429 (2015).
Ptriaz-TA	Humidity	8/9	---	0.32/---	Nat. Commun. <b>9</b> , 1717 (2018).
CNP	Humidity	---	---	---/120	Nat. Mater. <b>15</b> , 1084 (2016).

### Supplementary References

[S1]J. Li, J. Fan, R. Cao, Z. Zhen, J. Du et al., Encapsulated dye/polymer nanoparticles prepared via miniemulsion polymerization for inkjet printing. ACS Omega **3**(7), 7380-7387 (2018). <https://doi.org/10.1021/acsomega.8b01151>