

Supporting Information for

An Endotenon Sheath-Inspired Double-Network Binder Enables Superior Cycling Performance of Silicon Electrodes

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Supplementary Figures

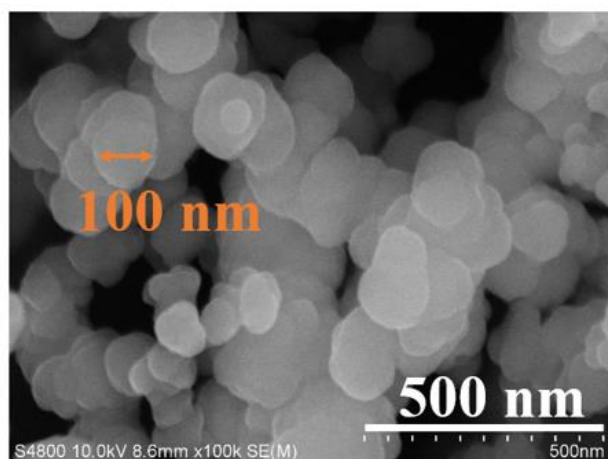


Fig. S1 Typical SEM image of silicon particles

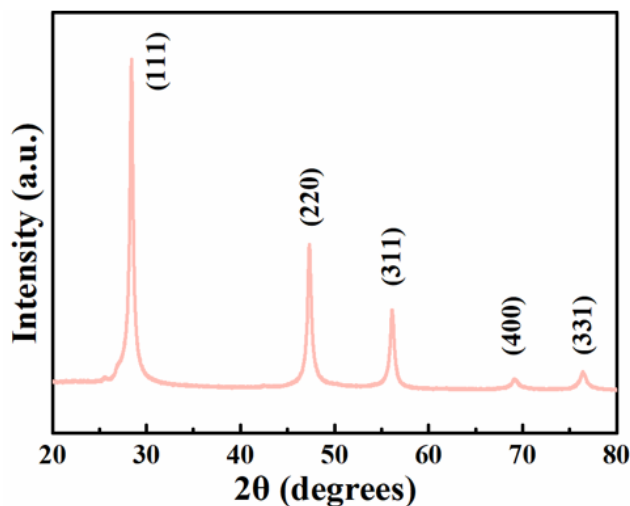


Fig. S2 XRD pattern of Si particles

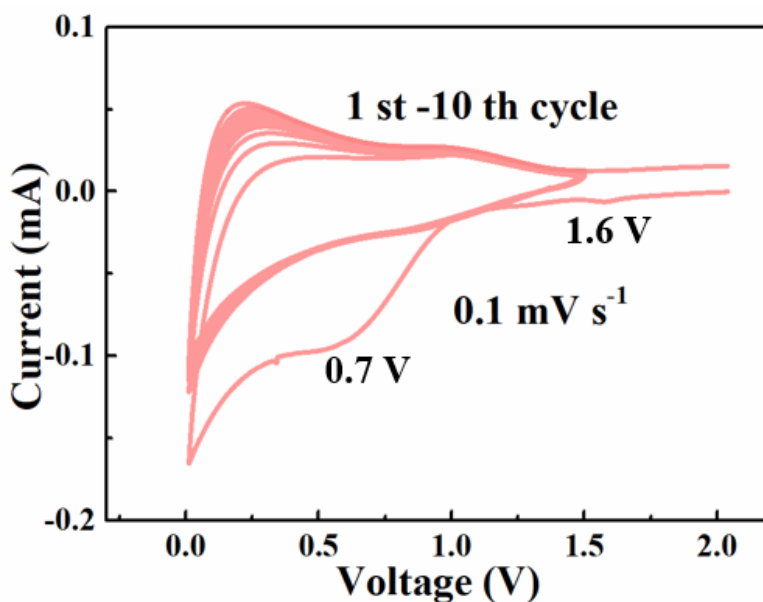


Fig. S3 CV curves of the DNB-based Si electrode at a scanning rate of 0.1 mV s^{-1}

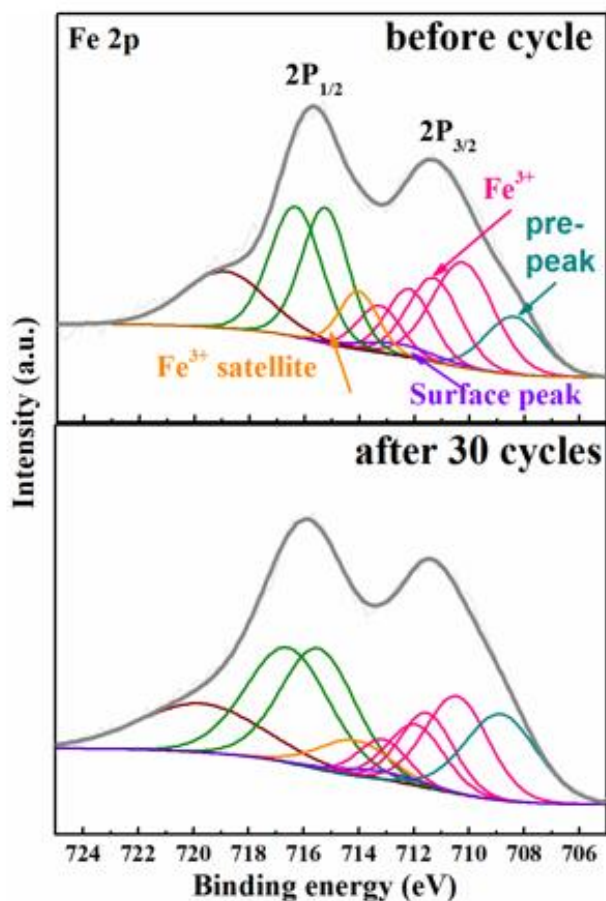


Fig. S4 Fe 2p XPS spectra with the DNB-based Si electrode before and after 30 cycles
 Fe 2p XPS spectra of DNB-based Si electrode before and after 30 cycles is shown in Fig. S4. It can be seen that after battery cycling, there is no obvious peak change in the $2p_{3/2}$ band of Fe^{3+} that consists of four subpeaks at the range of 710~714 eV, and meanwhile no peak in the range of 706~708 eV corresponding to reduction products of $\text{Fe}(0)$ or FeO . These findings indicate the good electrochemical stability of DNB.

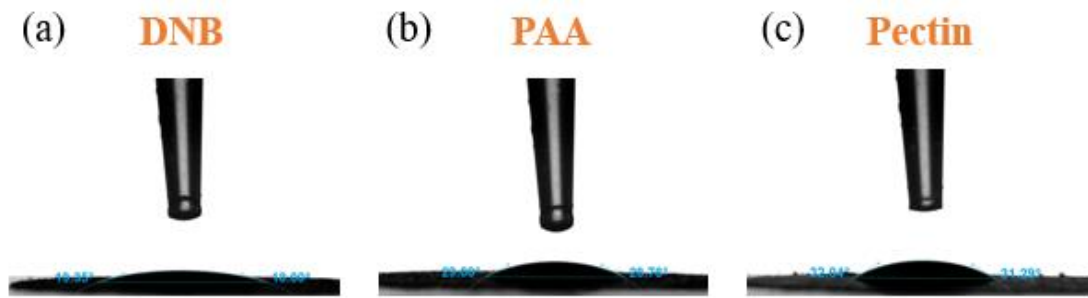


Fig. S5 Contact angles of $\text{LiPF}_6/\text{EC-DEC-FEC}$ electrolyte on (a) DNB film (19.35°), (b) PAA film (29.00°) and (c) pectin film (32.04°)

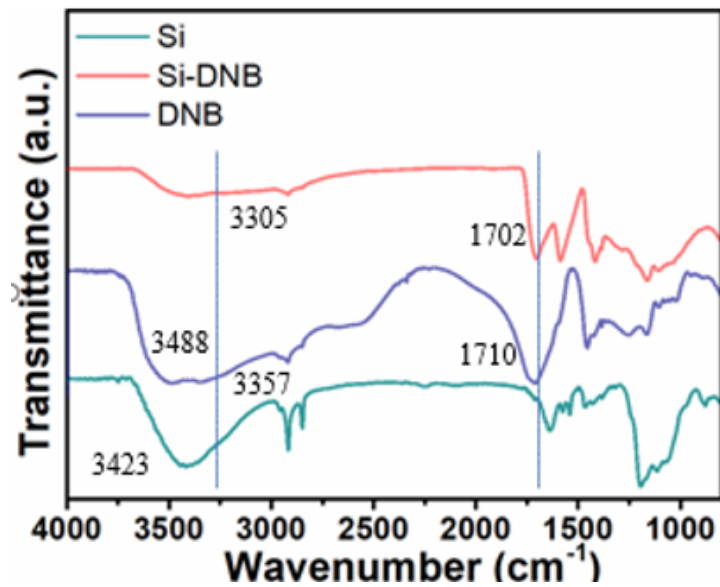


Fig. S6 FTIR spectra of Si, DNB and Si-DNB

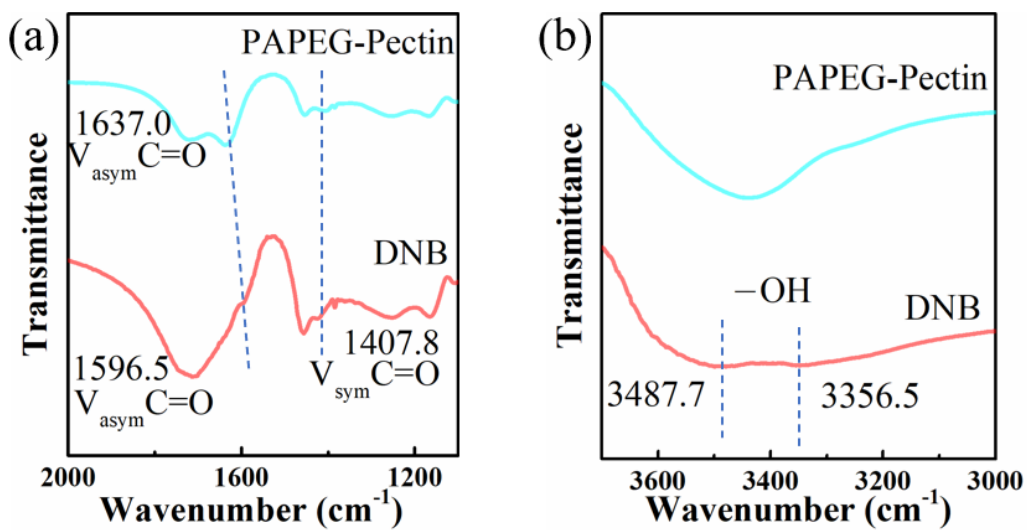


Fig. S7 FTIR spectra of DNB and PAPEG-pectin

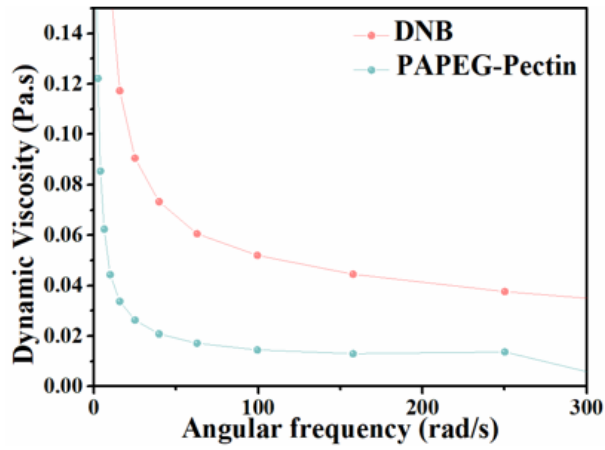


Fig. S8 Dynamic viscosity of DNB and PAPEG-pectin at different angular frequency and 17 °C

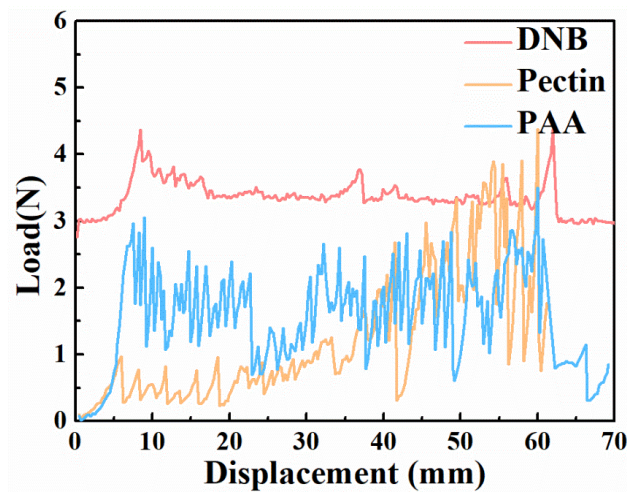


Fig. S9 Peel off test results of Si anodes based on DNB, PAA, and Pectin binders

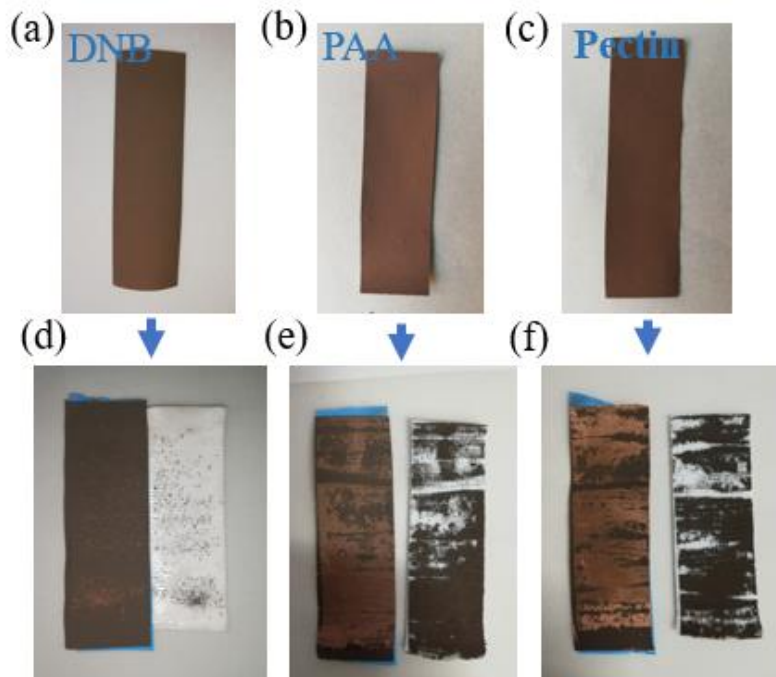


Fig. S10 Optical images of (a, d) DNB, (b, e) PAA and (c, f) pectin binder-based Si anodes before/after peeling test under the force of the tape

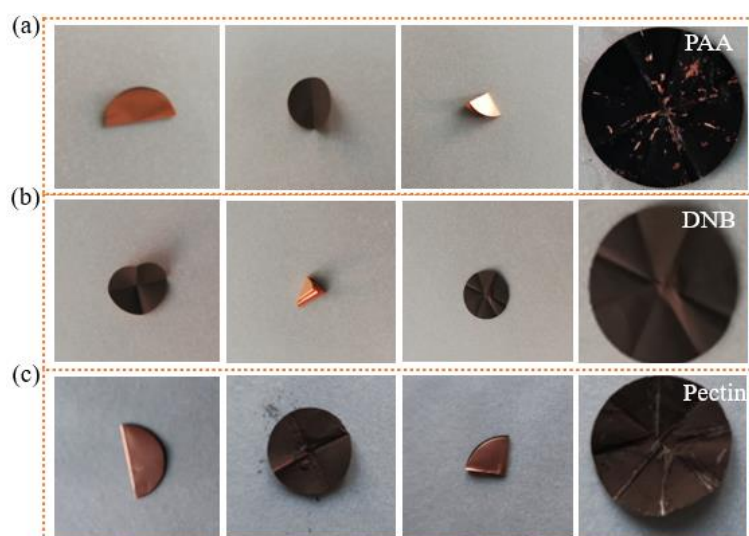


Fig. S11 Digital photographs of the (a) PAA, (b) DNB and (c) pectin-based Si electrode after being folded for one, two and three times, as well as their corresponding unfolded states

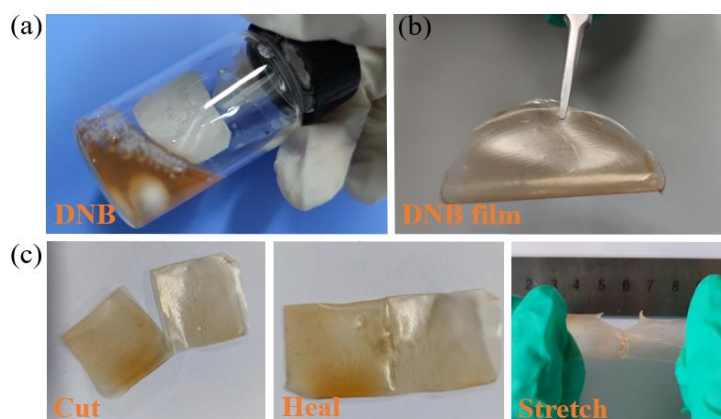


Fig. S12 The photographs of aqueous (a) DNB solution and (b) DNB film and (c) the self-healing ability of DNB

As shown in Fig. S12, the DNB film was prepared by coating the aqueous polymer-containing solution (5 wt%, Fig. S12a) on a Teflon mould at room temperature, followed by drying under a vacuum oven at 80 °C for 10 h. The dry polymer films were cut into 2 cm × 2cm for self-healing testing.

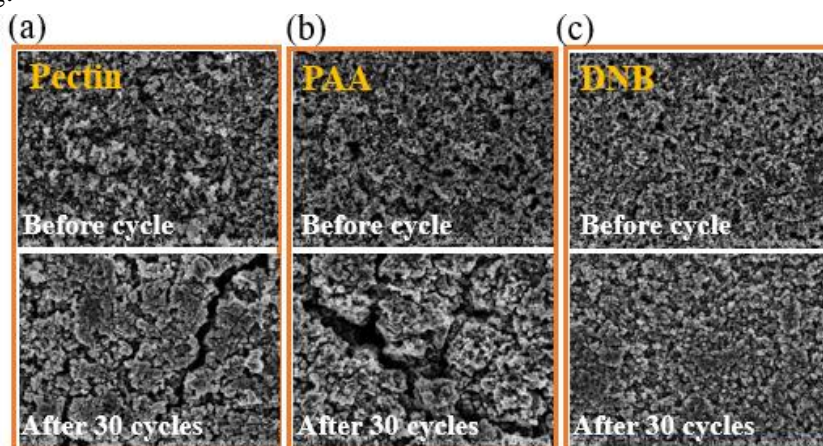


Fig. S13 Typical SEM images of (a) the pristine/the 30-cycled DNB-based Si electrode, (b) the pristine/the 30-cycled PAA-based Si electrode, (c) the pristine/the 30-cycled pectin-based Si electrode

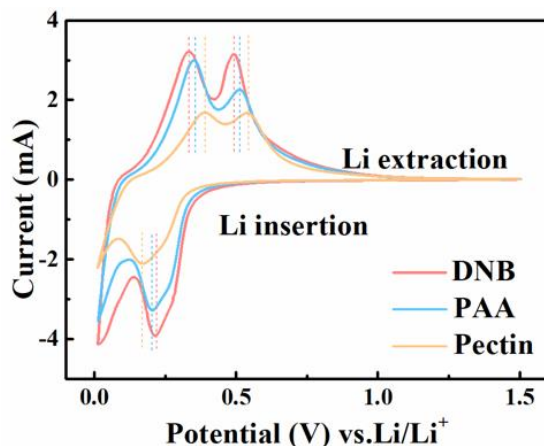


Fig. S14 CV curves of the different binders-based Si/Li cells at a scanning rate of 0.1 mV s^{-1} after five precycles

As illustrated in Fig. S14, the reduction peak at $\sim 0.2 \text{ V}$ is assigned to the reversible lithiation process of amorphous Si domains to Li-Si phase. The Si reduction peak at $\sim 0.01 \text{ V}$ is ascribed to the formation of amorphous Li-Si phases from the crystalline Si. In the delithiation process, two oxidation peaks at about 0.44 and 0.55 V are attributed to the delithiation process of Li-Si phases to amorphous Si domains. The higher peak current in the presence of the DNB indicates an overall increase in the dealloying and alloying kinetics.

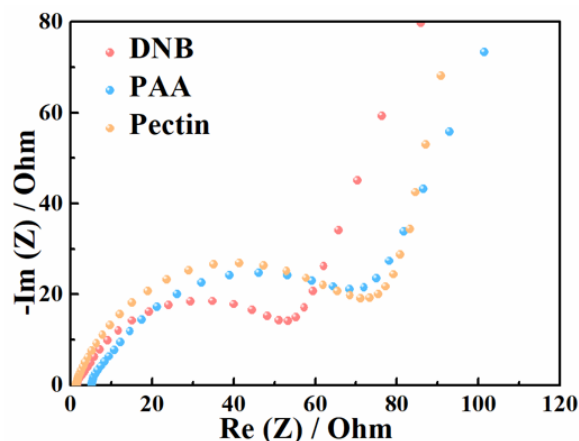


Fig. S15 Electrochemical impedance spectra of Si electrodes using DNB, as well as PAA and pectin binders after five precycles

As shown in Fig. S15, the DNB-based Si electrode has the smallest battery impedance (i.e., the sum of SEI impedance and charge-transfer resistance).

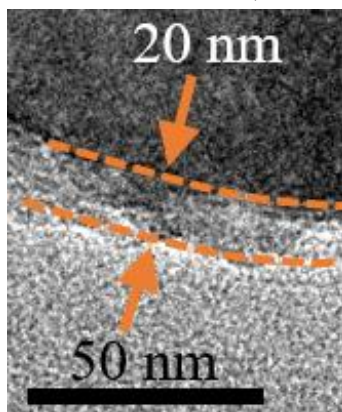


Fig. S16 Typical TEM image of the SEI on the Si surface with DNB after 30 cycles

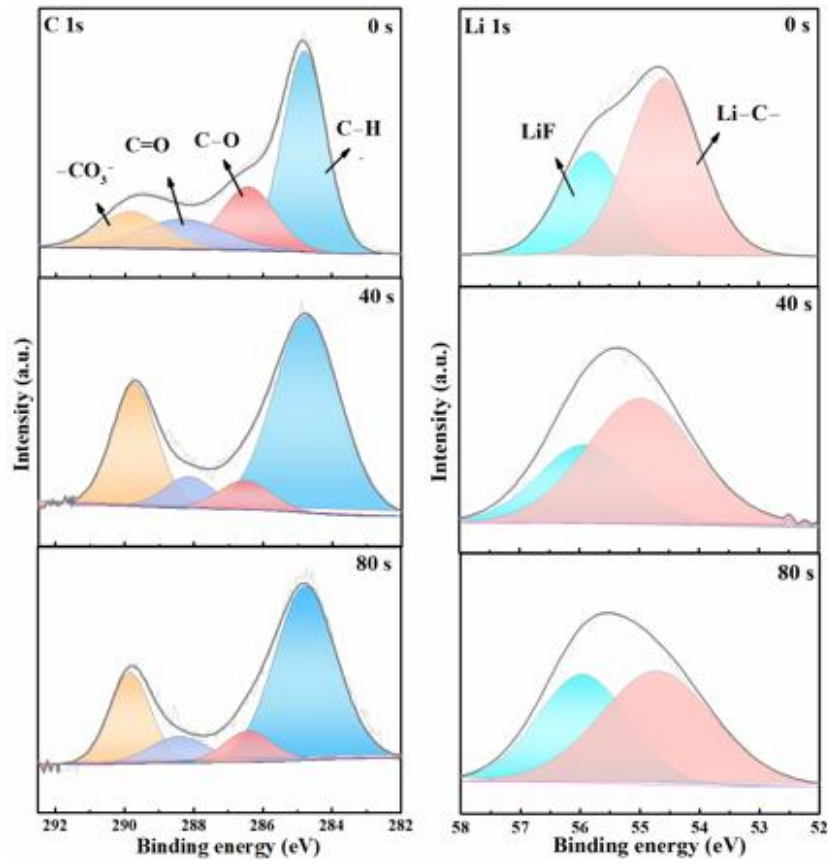


Fig. S17 (a) C 1s, (b) Li 1s XPS spectra of different Ar ion sputtering time with the PAA-based Si electrode after 30 cycles

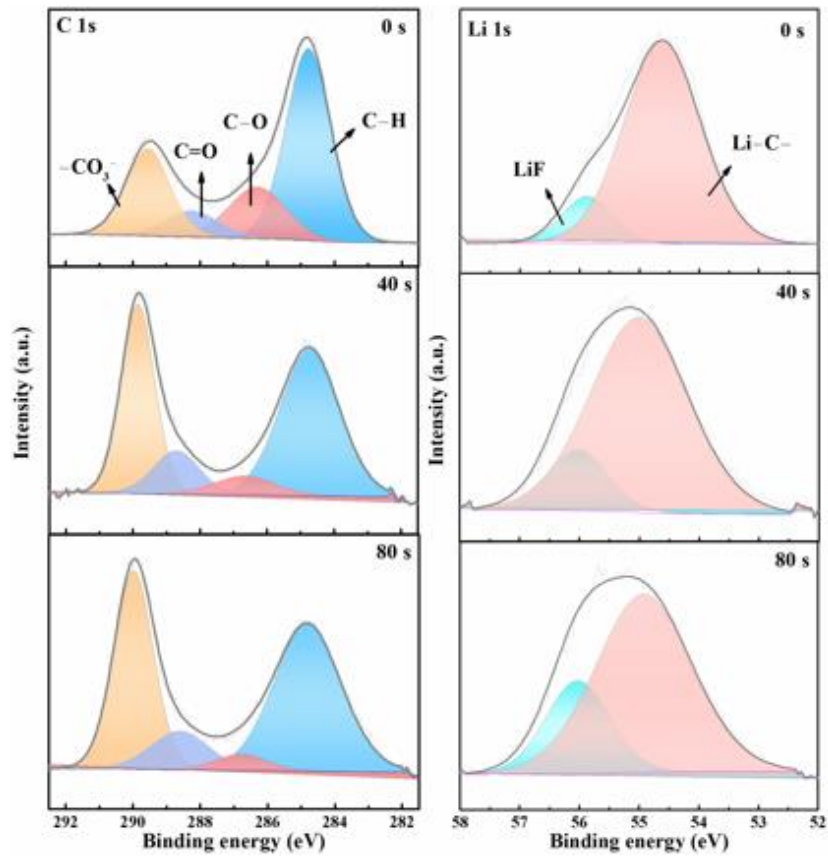


Fig. S18 (a) C 1s, (b) Li 1s XPS spectra of different Ar ion sputtering time with the pectin-based Si electrode after 30 cycles

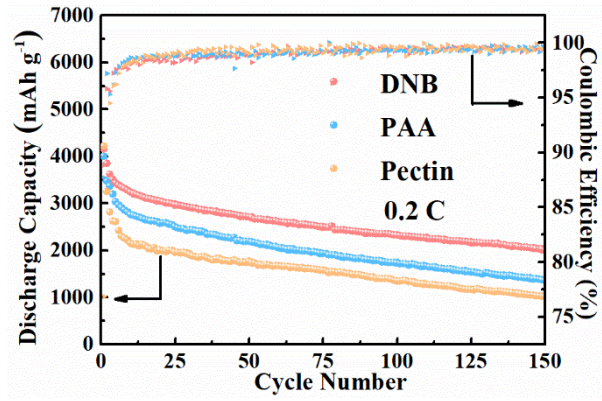


Fig. S19 Cycling performance and coulombic efficiency of the Si (0.93 mg/cm²)/Li cells with different binders at 0.2 C, room temperature and a voltage range of 0.01-1.5 V